

Physical and Chemical Changes

Identify if the actions depict a physical or chemical change.

11



cutting an apple

- ☐ physical change
- ☐ chemical change

12



baking a cookie

- ☐ physical change
- ☐ chemical change

13



burning wood

- ☐ physical change
- ☐ chemical change

14



ice melting

- ☐ physical change
- ☐ chemical change

15



rotting fruit

- ☐ physical change
- ☐ chemical change

16



digesting food

- ☐ physical change
- ☐ chemical change

17



boiling water

- ☐ physical change
- ☐ chemical change

18



chopping wood

- ☐ physical change
- ☐ chemical change

19



cracking an egg

- ☐ physical change
- ☐ chemical change

20



nailing nail

- ☐ physical change
- ☐ chemical change

Cphs Physical And Chemical Change Worksheet

J Spring



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Physical and Chemical Changes (ENHANCED eBook) Edward P. Ortleb, Richard Cadice, 1993-09-01 This book presents a program of basic studies in physical and chemical changes of matter. The definition of matter is presented along with explanations of states and properties of matter. Topics include atoms, molecules, elements, compounds, mixtures, solutions, symbols, and formulas. Each of the twelve teaching units in this book is introduced by a color transparency, print books, or PowerPoint slide eBooks that emphasize the basic concept of the unit and presents questions for discussion. Reproducible student pages provide reinforcement and follow-up activities. The teaching guide offers descriptions of the basic concepts to be presented, background information, suggestions for enrichment activities, and a complete answer key. Physical and Chemical Change, 1995

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Chemical Changes, 2003-01-14 Any change that results in the formation of a new kind of matter is a chemical change. Energy is always involved in a chemical change, sometimes with spectacular results. Chemical Changes, 2004 Any change that results in the formation of a new kind of matter is a chemical change. Energy is always involved in a chemical change, sometimes with spectacular results. *Science Workshop Series: Chemical changes* Seymour Rosen, 2000 This program presents science concepts in areas of biology, earth science, chemistry, and physical science in a logical, easy-to-follow design that challenges without overwhelming. This flexible program consists of 12 student texts that can easily supplement an existing science curriculum or be used as a stand-alone course. Reading Level 4-5, Interest Level 6-12. **Chemical Changes** Rebecca L. Johnson, 2008-06-01 Matter has both physical and chemical properties. During a chemical change, the chemical properties of a substance change and one or more new substances form. Every chemical reaction involves a transfer of energy. During a chemical reaction, atoms in matter are rearranged, but no atoms are gained or lost. **School Certificate Chemistry Form 3**, **Chemical Changes** Rebecca L. Johnson, 2008-06-01 Physical and chemical properties of matter can change. In a chemical reaction, reactants change into products. Energy is either absorbed or released, but matter is always conserved. Bonds in reactants are broken, and atoms are rearranged to form products. The pH scale measures the strength of acids and bases found in many substances. **Chemical Changes** Lynnette Brent, 2008-08 An introduction to how chemicals

react and change **Explaining Chemical Change : Student Exercises and Teacher's Guide for Grade Nine**

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