

Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON_METALS share electrons, attempting to attain a stable octet (8 outer electrons) in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

1.) H_2 hydrogen is diatomic:



2.) F_2 fluorine is diatomic:



3.) O_2 oxygen is diatomic:



4.) N_2 nitrogen is diatomic. Is this a triple bond?



5.) BF_3 you need 3 fluorine atoms here:



6.) Ammonia NH_3 hint: how many hydrogen atoms are needed?



7.) Carbon dioxide CO_2



8.) Methane CH_4 careful here 4 hydrogen atoms needed



9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!!



10.) SO_2 hint: one pair of electrons from sulfur must be slit up for this one to work.



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L Reisser



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Prentice Hall Science Explorer Michael J. Padilla, 2002 **Prentice Hall Chemistry**, 2000 *Prentice Hall Science Explorer: Chemical Interactions* Pearson/Prentice Hall, Michael J. Padilla, 2004-10 **Prentice Hall Biology** Sandra Gottfried, Gerry M. Madrazo, Jr., 1990 *Prentice Hall General Science: A voyage of discovery*, 1992 **Chemical Matter** Prentice-Hall Staff, 1994 Authorized teaching resource in Alberta for senior high science 14 24 1995 2004 **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *Biology* Kenneth Raymond Miller, Prentice Hall (School Division), 1999-02

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