

2005

 **TOYOTA**



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VOLUME 1

2005

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exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of
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eg 1 hcl aq + naoh aq → nacl aq + h₂o l for which ΔH = 57 kJ mol⁻¹ eg 2 h₂no₃ aq + koh aq → kno₃ aq + h₂o l for which ΔH = 57 kJ mol⁻¹

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