Strengths of Acids and Bases

- Strengths of Acids
 - Strong acid acid that ionizes completely
 - ☐ Produce the maximum number of ions
 - □ Good conductors of electricity
 - □ Ex: hydrochloric acid (HCl)
 - Weak acid acid that ionizes only partially
 - □ Some of the acid remains in molecular form
 - □ Ex: acetic acid (vinegar)
 - Using the Bronsted-Lowry model, if the reactant acid is strong, its conjugate base is weak

19 3 Chemistry Answers Strengths Of Acids And Bases

Mr. Ajeet Singh

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