| Name | Date | Class |
|--------------------------|--|-------|
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7-3 Review and Reinforcement

Naming Chemical Compounds

Write the correct numerical prefix for each of the numbers in the table below.

| Number | Prefix | | | | |
|--------|--------|------------------------------|--|--|--|
| 1 | mono | | | | |
| 2 | di | | | | |
| 3 | tri | | | | |
| 4 | tetra | | | | |
| 5 | penta | | | | |
| 6 | hexa | to an analysis of the second | | | |
| 7 | hepta | 1-01/17 0 9 | | | |
| 8 | octa | | | | |
| 9 | nona | | | | |
| 10 | deca | | | | |

Complete each of the following sentences by filling in the appropriate word or phrase from the list below

cation hydrates electronegative ionic anion anhydrates hydrogen ions molecular

11. Ionic compounds that absorb water into their solid structures form substances called

hydrate —

- 12. When naming an ionic substance, the name of the __cation _____ comes first
- To name a binary molecular compound, add the suffix -ide to the more_electronegative
 of the two elements.
- 14. By one definition, an acid is a molecular compound that dissolves in water to produce hydrogen ions
- 15. The name of an acid is usually derived from the name of its characteristic anion
- 16. Prefixes are often used when naming molecular compounds.

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Diana L. Turner

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- 1. Understanding the eBook 22 2 Review And Reinforcement Chemistry Answers
 - The Rise of Digital Reading 22 2 Review And Reinforcement Chemistry Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying 22 2 Review And Reinforcement Chemistry Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 22 2 Review And Reinforcement Chemistry Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from 22 2 Review And Reinforcement Chemistry Answers
 - Personalized Recommendations

- 22 2 Review And Reinforcement Chemistry Answers User Reviews and Ratings
- 22 2 Review And Reinforcement Chemistry Answers and Bestseller Lists
- 5. Accessing 22 2 Review And Reinforcement Chemistry Answers Free and Paid eBooks
 - 22 2 Review And Reinforcement Chemistry Answers Public Domain eBooks
 - 22 2 Review And Reinforcement Chemistry Answers eBook Subscription Services
 - 22 2 Review And Reinforcement Chemistry Answers Budget-Friendly Options
- 6. Navigating 22 2 Review And Reinforcement Chemistry Answers eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - 22 2 Review And Reinforcement Chemistry Answers Compatibility with Devices
 - 22 2 Review And Reinforcement Chemistry Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of 22 2 Review And Reinforcement Chemistry Answers
 - Highlighting and Note-Taking 22 2 Review And Reinforcement Chemistry Answers
 - Interactive Elements 22 2 Review And Reinforcement Chemistry Answers
- 8. Staying Engaged with 22 2 Review And Reinforcement Chemistry Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 22 2 Review And Reinforcement Chemistry Answers
- 9. Balancing eBooks and Physical Books 22 2 Review And Reinforcement Chemistry Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 22 2 Review And Reinforcement Chemistry Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 22 2 Review And Reinforcement Chemistry Answers
 - Setting Reading Goals 22 2 Review And Reinforcement Chemistry Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 22 2 Review And Reinforcement Chemistry Answers
 - Fact-Checking eBook Content of 22 2 Review And Reinforcement Chemistry Answers

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