Acid - Base Titration Lab

Knowledge	/16
Introduction (Knowledge component) -acids and bases are well defined -strong and weak acid/base is clearly explained -neutralization is well explained -good definition of titration and explanation -relevant real-world application of acid-base chemistry described -purpose of the lab	/16 0 1 2 3 0 1 2 3 0 1 2 3 0 1 2 3 0 1 2 3 0 1 2 3
Communication	/12
Title Page -Includes appropriate and descriptive title for the lab	/1 0 1
Introduction (Communication component) -written in full paragraphs (no headings/questions) -facts are correctly cited	/4 0 1 2 0 1 2
Results - Observation Tables - Appropriate titles at the top - Units are in the headings - Well organized and neat (ex. numbers are aligned) - Contains all relevant data and observations (including averages)	/5 0 1 0 1 0 1 0 1 2
Overall clarity, neatness, formatting, spelling and grammar	/2
Inquiry	/17
Analysis -Balanced chemical reaction is complete -Full calculations showing how to determine the concentration of HCI	/4 0 1 0 1 2 3
Discussion -difference between end point and equivalence point -NaOH's hygroscopic nature and its effect on the results -Two experimental sources of error discussed and explained	/9 0 1 2 0 1 2 3 0 1 2 3 4
Conclusion -Final statement summarizing results of the lab -One thing that was learned from the lab	/2 0 1 0 1
References/Work Cited -Correctly written in APA format	/2

Total: /45

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