

1. As a solute is added to a solvent, what happens to the freezing point and the boiling point of the solution?
- ☒ A) The freezing point decreases and the boiling point increases.
- B) The freezing point increases and the boiling point increases.
- C) The freezing point increases and the boiling point decreases.
- D) The freezing point decreases and the boiling point decreases.
2. Compared to the freezing point of 1.0 M KCl(aq) at standard pressure, the freezing point of 1.0 M CaCl₂(aq) at standard pressure is
- More ions* ☒ A) lower
- B) higher
- C) the same
3. Which aqueous solution of KI freezes at the lowest temperature?
- A) 1 mol of KI in 500. g of water
- ☒ B) 2 mol of KI in 500. g of water *(more concentrated)*
- C) 1 mol of KI in 1000. g of water
- D) 2 mol of KI in 1000. g of water
4. A 1 kilogram sample of water will have the highest freezing point when it contains
- A) 1×10^{21} dissolved particles
- ☒ B) 1×10^{17} dissolved particles
- C) 1×10^{19} dissolved particles
- D) 1×10^{23} dissolved particles *lowest f.p.*
5. A solution consists of 0.50 mole of CaCl₂ dissolved in 100. grams of H₂O at 25°C. Compared to the boiling point and freezing point of 100. grams of H₂O at standard pressure, the solution at standard pressure has
- ☒ A) a higher boiling point and a lower freezing point
- B) a lower boiling point and a lower freezing point
- C) a higher boiling point and a higher freezing point
- D) a lower boiling point and a higher freezing point
6. Which solution will freeze at the lowest temperature?
- A) 1 mole of sugar in 500 g of water
- B) 1 mole of sugar in 1,000 g of water
- C) 2 moles of sugar in 1,000 g of water
- ☒ D) 2 moles of sugar in 500 g of water
7. Which solution has the highest boiling point at standard pressure?
- 4 ions* ☒ A) 0.10 M K₃PO₄(aq) *3 K⁺ 3 PO₄⁻*
- B) 0.10 M KCl(aq)
- C) 0.10 M KNO₃(aq)
- D) 0.10 M K₂SO₄(aq)
8. Which concentration of a solution of CH₃OH in water has the lowest freezing point?
- A) 0.001 M
- ☒ B) 0.1 M
- C) 0.01 M
- D) 0.0001 M
9. How do the boiling point and freezing point of a solution of water and calcium chloride at standard pressure compare to the boiling point and freezing point of water at standard pressure?
- A) Both the freezing point and boiling point of the solution are higher.
- B) The freezing point of the solution is higher and the boiling point of the solution is lower.
- ☒ C) The freezing point of the solution is lower and the boiling point of the solution is higher.
- D) Both the freezing point and boiling point of the solution are lower.
10. Compared to a 2.0 M aqueous solution of NaCl at 1 atmosphere, a 3.0 M aqueous solution of NaCl at 1 atmosphere has a
- A) lower boiling point and a higher freezing point
- B) higher boiling point and a higher freezing point
- C) lower boiling point and a lower freezing point
- ☒ D) higher boiling point and a lower freezing point

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web für die meisten kunststoffe folgt der synthese bei den rohstoffherstellern die davon völlig getrennte verarbeitung dies bedeutet dass abkühlung und erneute aufheizung sowie transport manchmal mehrfach dazwischen liegen ausnahmen bilden bisher

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web als kunststoffverarbeitung wird die herstellung von kunststoffprodukten bezeichnet dies geschieht insbesondere durch die fertigung von formteilen halbzeugen fasern oder folien aus kunststoffen die von der chemischen industrie meist als granulat masterbatch pulver folien oder platten geliefert werden

grundlagen der kunststoffverarbeitung springerlink - May 13 2023

web oct 15 2013 in diesem Spannungsfeld zwischen Form, Material und Verarbeitungstechnik gehören daher auch Kenntnisse der Kunststoffverarbeitung zum Grundwissen eines Materialwissenschaftlers. Das folgende Kapitel wird daher einen kurzen einführenden Überblick in die wesentlichen Grundbegriffe der Kunststoffverarbeitung geben.

Verarbeitung von Kunststoffen Methoden Studysmarter - Aug 16 2023

Web: Verarbeitung von Kunststoffen. Das Wichtigste: Kunststoffe bestehen aus Polymeren und können in linearer, verzweigter und vernetzter Form vorkommen. Es gibt drei Arten von Kunststoffen: Duroplaste, Elastomere und Thermoplaste. Das erste Verarbeiten von Kunststoff in Form von Granulat oder Pulver heißt Urformen.