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(a) The symbol equation for the reaction can be represented as shown in Fig. 3.1.

Fig. 3.1

Table 3.1 shows some bond energies.

| bond                  | N-H | N-N |
|-----------------------|-----|-----|
| bond energy in kJ/mol | 435 | 165 |

Use the bond energies in Table 3.1 and  $\Delta H$  to calculate the bond energy of an **N-O bond** in kJ/mol.

Use the following steps.

- Calculate the energy needed to break bonds in the reactants.  

$$(4 \times 435) + 1(165)$$

$$4 \times 435 + 1(165)$$

$$2175 \text{ kJ}$$
- Calculate the energy released when bonds form in the products.  

$$\text{bond broken} - \text{bond formed} = -90$$

$$2175 - x = -90$$

$$x = 2175 + 90$$

$$2265 \text{ kJ}$$
- Calculate the energy of **an N-O bond**.  

$$6 \times (\text{N-O}) = 2265$$

$$\text{N-O} = \frac{2265}{6}$$

$$377.5 \text{ kJ/mol}$$

# June 14 0620 Paper 21

**Geological Survey (U.S.)**



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web en bref nous pouvons voir qu'il y a différents aspects du bouddhisme que les gens trouvent attirants et intéressants mais tout cela vient du souhait fondamental de trouver

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pour plus de détails on se référera aux articles détaillés par pays liens dans le tableau de statistiques ci-dessous bouddhisme en Birmanie on recense environ 100 000 moines birmans la pratique bouddhique coexiste avec le culte des nats bouddhisme au Cambodge depuis 1993 le bouddhisme est religion d'État le Cambodge n

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web le bouddha a aussi donné l'instruction à ses disciples de ne pas boire même une goutte d'alcool les pratiques du bouddhisme visent à cultiver l'attention la discipline et le