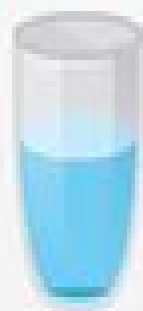


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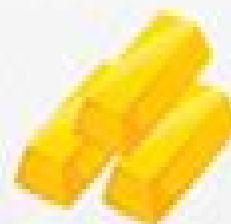
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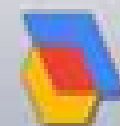
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Which Is A Pure Substance Edgenuity

**Karl H. Simmrock, Ralf
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Pure Substances. Part 2 _ Compounds from BeBr_g to ZrCl2_g Scientific Group Thermodata Europe (SGTE),1999-10-20 Thermodynamic data for inorganic materials are fundamental for the optimisation of existing process parameters and for investigating suitable parameters for carrying out potential new processes With the aid of such data time and costs can be saved by calculating the conditions necessary to produce a material of the required composition and specified purity with a minimum usage of energy and input materials and with a minimum release of harmful substances to the environment The SGTE evaluated data presented here are tabulated values of standard thermodynamic properties enthalpy of formation and standard entropy at 298 15K enthalpies and temperatures of transition heat content for each substance together with plotted heat capacity Gibbs energy and enthalpy of formation functions up to the maximum temperature for which the data for that substance have been evaluated The data are presented in 3 subvolumes A Pure Substances B Binary Systems C Ternary and Multi Component Systems *Pure Substances. Part 3 _ Compounds from CoCl3_g to Ge3N4* Scientific Group Thermodata Europe (SGTE),2000-08-28 Thermodynamic data for inorganic materials are fundamental for the optimisation of existing process parameters and for investigating suitable parameters for carrying out potential new processes With the aid of such data time and costs can be saved by calculating the conditions necessary to produce a material of the required composition and specified purity with minimum usage of energy and input materials and with minimum release of harmful substances to the environment The SGTE evaluated data presented here are tabulated values of standard thermodynamic properties enthalpy of formation and standard entropy at 298 15K enthalpies and temperatures of transition heat content for each substance together with plotted heat capacity Gibbs energy and enthalpy of formation functions up to the maximum temperature for which the data for that substance have been evaluated The data are presented in 3 subvolumes A Pure Substances B Binary Systems C Ternary and Multi Component Systems **Critical Data of Pure Substances** Karl H. Simmrock,Ralf Janowsky,Andreas Ohnsorge,1986

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