

## EXERCISE - AQUEOUS SOLUTIONS AND WATER QUALITY

As you read through pages 357-362, answer the following questions:

- 1) What % of the world's fresh water does Canada have?
- 2) Look at the data in Table 9.5. Did you realize that it is acceptable to have small amounts of substances such as uranium, mercury and insecticides in drinking water?
- 3) a) What type of bacteria in the water supply caused 7 people to die and many to become ill in Walkerton, Ont. in 2000?  
b) What should the town's water manager have done to deal with this problem?

### Sources That Compromise Water Quality

- 1) State three examples of point sources of pollution.
- 2) State one example of a diffuse source of pollution.
- 3) State one example of an indirect source of pollution.

### Treating Water for Your Home

- 1) Briefly describe the six processes that take place at a water treatment plant.
- 2) Why do some municipalities add sodium fluoride to drinking water?

### Hard Water and Soft Water/Treating Water at Home

- 1) Distinguish between hard water and soft water.
- 2) What is the most common type of rock that causes hard water?
- 3) Where is boiler scale commonly found and what problems does it cause?
- 4) a) In an ion exchange water softener, what ion exchanges with the calcium and other hardness causing ions in the water?  
b) Why is this kind of water softener not the best choice for some people?

### Waste-Water Treatment

What does each of the following mainly involve:

- 1) primary waste-water treatment?
- 2) secondary waste-water treatment?
- 3) tertiary waste-water treatment?

### Solution Stoichiometry

Read The Concentration of Ions Sample Problem on pages 348-349.  
Then, do question 8 on page 352.  
(You can check your answer to this question on page 367.)

# Water And Aqueous Systems Vocabulary Review Answer Key

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## Water And Aqueous Systems Vocabulary Review Answer Key:

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