

1. CLASSIFICATION OF MATTER

- Matter can exist in three physical states: solid, liquid and gas.



- Various properties of matter are used to differentiate between the two.
- Matter is classified into pure and impure.
- Homogeneous mixture: composition is uniform throughout.
- Heterogeneous mixture: composition is not uniform.
- Colloid: particles of substance are dispersed in another substance.
- Suspension: particles of substance are dispersed in another substance.

2. PROPERTIES OF MATTER AND THEIR MEASUREMENT

- Properties of matter are physical and chemical properties.
- Physical properties are those which can be measured without changing the identity of the substance.
- Chemical properties are those which can be measured by changing the identity of the substance.
- Colour, odour, melting point, density, etc. are some physical properties.
- Combustion, corrosion, etc. are some chemical properties.
- Quantitative measurement of physical properties is necessary for scientific study.
- Measurement of mass, length, time, etc. are some physical properties.
- Measurement of temperature, etc. are some physical properties.
- Measurement of mass, length, time, etc. are some physical properties.

UNIT CONVERSIONS AND DIMENSIONS

1 kg = 1000 g	1 t = 1000 kg	1 g = 1000 mg	1 mg = 1000 µg	1 µg = 1000 ng
1 m = 100 cm	1 km = 1000 m	1 cm = 10 mm	1 mm = 1000 µm	1 µm = 1000 nm

UNIT CONVERSIONS

- 1 kg = 1000 g, 1 t = 1000 kg, 1 g = 1000 mg, 1 mg = 1000 µg, 1 µg = 1000 ng

3. SIGNIFICANT FIGURES

Significant figures are those digits in a number which are known to be reliable. They are used to express the precision of a measurement.

Significant figures are those digits which are known to be reliable. They are used to express the precision of a measurement.

Rules for determining the number of significant figures:

- Non-zero digits are significant.
- Zeros between non-zero digits are significant.
- Zeros at the end of a number are significant if they are on the right side of a decimal point.
- Leading zeros are not significant.
- In a number written in scientific notation, all digits are significant.
- Exact values are those which are known to be true.
- Counting is the agreement of a particular value to the true value of the count.
- In addition and subtraction of significant figures, the result should have the same number of significant figures as the least number of significant figures in the original numbers.
- In multiplication and division of significant figures, the result should have the same number of significant figures as the least number of significant figures in the original numbers.

Aakash Chemistry Lab Manual

G Thomas



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