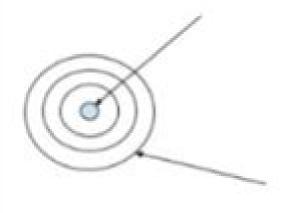
# **Atomic structure**

- 10			
- 60			
-85	440	ш	100
- 67	1000		

# Complete the following paragraph:

Atoms consist of a	composed of protons and	
The nucleus is surrounded by	which orbit in	-
Atoms have no net overall charge bed	and electrons	
balance and their	cancel each other out.	

# Label the diagram and complete the table below:



Subatomic particle	Relative mass	Relative charge
Proton		
Neutron		
Electron		

Atomic number = number of	
Mass number = number of	+ number of

He-=

so in helium, the number of protons = the atomic number (2), the number of

# **Atomic Structure Chemetry Workbook Answers**

**Arshad Iqbal** 

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