Nuclear Equation Worksheet



h)
$$_{00}^{234}$$
Pa \rightarrow + $_{10}^{0}$ e

i)
$${}^{234}_{91}$$
Pa \rightarrow + ${}^{4}_{2}$ He

k)
$$\rightarrow {}^{4}_{2}He + {}^{208}_{81}\Pi$$

$$() \stackrel{239}{\circ_4} Pu \rightarrow + {}^{4}_{2} He$$

2. For problem 1, label each reaction as involving an alpha, beta, or gamma particle. Also, label whether it is an example of decay (particle being given off) or capture (particle being taken in).

Write down the following equations.

a) Alpha decay of Curium - 247

b) Beta capture of manganese - 53

c) Beta decay of sulfur - 35

d) Beta capture of tellurium - 121

d) Gamma radiation given off by carbon - 13

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Michal Rosen-Zvi

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