

Test Tube S:

	$\text{Fe}^{3+} (\text{aq})$	+	$\text{SCN}^{-} (\text{aq})$	$\rightarrow$	$\text{FeSCN}^{2+} (\text{aq})$
I	$0.2 \text{ M} (18 \text{ mL} / 20 \text{ mL})$ $= 0.18 \text{ M}$		$0.002 \text{ M} (2 \text{ mL} / 20 \text{ mL})$ $= 0.0002 \text{ M}$		0
C	-0.0002 M		-0.0002 M		$+0.0002 \text{ M}$
E	Large Excess		0		0.0002 M

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Asa Determination Of The Equilibrium Constant

Jianjun Gao



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