

## Applications of Aqueous Equilibria– ch. 8

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14. Will  $\text{BaCrO}_4$  ( $K_{\text{sp}} = 8.5 \times 10^{-11}$ ) precipitate when 200 mL of  $1 \times 10^{-5}$  M  $\text{Ba}(\text{NO}_3)_2$  is mixed with 350 mL of  $3 \times 10^{-5}$  M  $\text{KCrO}_4$ ?

# Applications Of Aqueous Equilibria Answer Key

**H Kauffman**



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