



Defence

Department
Defence

REPUBLIC OF SOUTH AFRICA



SOUTH AFRICAN NAVY

"THE PEOPLES NAVY"



COMBAT, ENGINEERING, TECHNICAL, SUPPORT AND MARITIME REACTION SQUADRON (MRS) APPLICANTS 2019

BACKGROUND

MILITARY SKILLS DEVELOPMENT SYSTEM (MSDS)

The SA Navy is offering young South African citizens an excellent opportunity to serve in uniform over a two-year period. The Military Skills Development System (MSDS) is a two-year voluntary service system with the aim of equipping and developing young South Africans with the necessary skills in order to compete in the open market.

Young recruits are required to sign up for a period of two-years, during which they will receive Military Training and further Functional Training.

All applicants must comply with the following:

- South African Citizen (Relinquish any other citizenship).
- Have no record of criminal offences.
- Not be under bond.
- Preferably single.
- Comply with medical fitness requirements for appointments in the SANDF.
- Be free from physical disabilities and not suffer from claustrophobia or a fear of heights.

Minimum Requirements for Applicants with Matric/Grade 12 only:

- Combat applicants aged between 18 and 22 must have completed Grade 12 with Mathematics and Physical Science with at least Level 3 for both subjects.
- Engineering and Technical applicants aged between 18 and 22 must have completed Grade 12 with Mathematics and Physical Science with at least Level 3 for both subjects.
- MRS/Protection Personnel applicants aged between 18 and 22 must have completed Grade 12.

Requirements for Graduates:

- Graduate applicants aged between 18 and 26 must have a completed Grade 12 with a Degree, National Diploma or N6 with Trade Test Certificate in Mechanical, Marine or Electrical Engineering.
- Graduate applicants aged between 18 and 26 must have completed Grade 12 with a Degree or National Diploma in Human Resource or Logistics.

THIS APPLICATION FORM MUST BE SUBMITTED WITH THE FOLLOWING CERTIFIED DOCUMENTS:

1. Copy of ID
2. Copy of Matric Certificate
3. A Short CV
4. Highest achieved qualification (e.g. Degree, National Diploma or N6 with Trade Test)

ENQUIRIES AND APPLICATIONS

Admission Enquiries can be directed to: SA Navy Recruitment Centre at (012) 329-4425/44252/45054. If interested in the MSDS in the SA Navy, kindly complete the application form with the requested documents & post to: Department of Defence, SA Navy Headquarters (Recruitment Centre) Private Bag X104, Pretoria, 0001

Closing Date: 28 February 2019

Should you not receive a response by 31 October 2018, kindly consider your application as unsuccessful.

Further information can be obtained on www.dod.mil.za or www.navy.mil.za

THE DEPARTMENT OF DEFENCE RESERVES THE RIGHT TO EMPLOYMENT

Members busy with Grade 12 need not apply

THIS APPLICATION FORM IS NOT FOR SALE

MILITARY SKILLS DEVELOPMENT SYSTEM: 2019

SCARCE SKILLS MUSTERING

Matric
Degree

☐

National Diploma
N6 with Trade Test

☐

Please complete the following Biographical Information:

First Names:

Surname:

ID Number:

Tel.(H):

Tel.(W):

Cell No.

Parents Cell No.

I will be years of age on 1 January 2019

Postal Address:

Province and nearest major town/city:

Gender:

Female

☐

Male

☐

Race:

☐☐☐☐☐

Grade 12:

Tertiary:

I hereby apply for

Subject, Symbols and Levels:

Subject	Symbol	Grade	Level

Applicants will be subjected to Psychometric Testing, Medical Evaluation, Vetting on criminal record and Verification of qualifications during the selection process.

Signature of Applicant: Date 2018.

This Application form will serve as a document for pre-selection.

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Table of Contents Application Forms In Sa Navy

1. Understanding the eBook Application Forms In Sa Navy
 - The Rise of Digital Reading Application Forms In Sa Navy
 - Advantages of eBooks Over Traditional Books
2. Identifying Application Forms In Sa Navy
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Application Forms In Sa Navy
 - User-Friendly Interface
4. Exploring eBook Recommendations from Application Forms In Sa Navy

- Personalized Recommendations
- Application Forms In Sa Navy User Reviews and Ratings
- Application Forms In Sa Navy and Bestseller Lists
- 5. Accessing Application Forms In Sa Navy Free and Paid eBooks
 - Application Forms In Sa Navy Public Domain eBooks
 - Application Forms In Sa Navy eBook Subscription Services
 - Application Forms In Sa Navy Budget-Friendly Options
- 6. Navigating Application Forms In Sa Navy eBook Formats
 - ePub, PDF, MOBI, and More
 - Application Forms In Sa Navy Compatibility with Devices
 - Application Forms In Sa Navy Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Application Forms In Sa Navy
 - Highlighting and Note-Taking Application Forms In Sa Navy
 - Interactive Elements Application Forms In Sa Navy
- 8. Staying Engaged with Application Forms In Sa Navy
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Application Forms In Sa Navy
- 9. Balancing eBooks and Physical Books Application Forms In Sa Navy
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Application Forms In Sa Navy
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Application Forms In Sa Navy
 - Setting Reading Goals Application Forms In Sa Navy
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Application Forms In Sa Navy

- Fact-Checking eBook Content of Application Forms In Sa Navy
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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web elements in the second row of the periodic table place their electrons in the 2n shell as well as the 1n shell for instance

lithium li has three electrons two fill the 1 s orbital and the third is placed in the 2 s orbital giving an electron configuration of 1 s 2 2 s 1

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web one more point needs to be emphasized about the relationship between electron configuration and the periodic table the atoms of elements in the same vertical column of the table have similar electron configurations for example consider the alkaline earth elements group iia

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web the electron removed during the ionization of beryllium he 2s² is an s electron whereas the electron removed during the ionization of boron he 2s² 2p¹ is a p electron this results in a lower first ionization energy for boron even though its nuclear charge is greater by one proton

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each period written out these are he 2 helium 1s² ne 10 neon 1s² 2s² 2p⁶

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web q1 what is the electron configuration of the element potassium answer the atomic number of potassium is 19 a neutral potassium element has 19 electrons its electron configuration is 1s² 2s² 2p⁶ 3s² 3p⁶ 4s¹ or simply put ar 4s¹ q2 what is the total number of electrons that can be filled in s p and d subshells

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