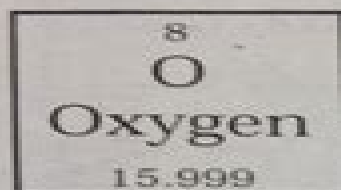


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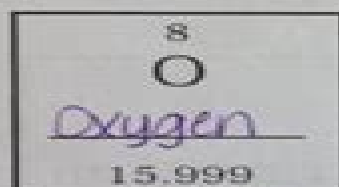
← Atomic Number
 ← Symbol
 ← Name
 ← Atomic Mass

Atomic number equals
the number of

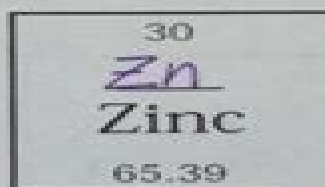
protons or electrons

Atomic mass equals
the number of

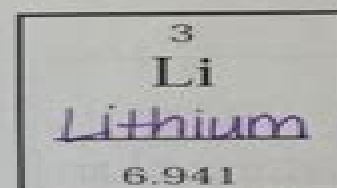
protons + neutrons



Atomic # = 8
 Atomic Mass = 16
 # of Protons = 8
 # of Neutrons = 8
 # of Electrons = 8



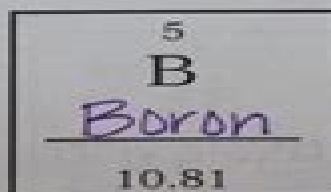
Atomic # = 30
 Atomic Mass = 65
 # of Protons = 30
 # of Neutrons = 35
 # of Electrons = 30



Atomic # = 3
 Atomic Mass = 7
 # of Protons = 3
 # of Neutrons = 4
 # of Electrons = 3



Atomic # = 14
 Atomic Mass = 28
 # of Protons = 14
 # of Neutrons = 14
 # of Electrons = 14



Atomic # = 5
 Atomic Mass = 11
 # of Protons = 5
 # of Neutrons = 6
 # of Electrons = 5



Atomic # = 35
 Atomic Mass = 80
 # of Protons = 35
 # of Neutrons = 45
 # of Electrons = 35

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H Kauffman



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