

Atoms and Atomic Structure (Worksheet)

Atomic and Atomic Structure (Worksheet)

Q1. Use the words in the box below to label the parts of the atom.

- Electron Proton
Neutron Nucleus



Q2. Which of the following statement is false?

- (a) Matter is composed of atoms.
(b) Protons and neutrons are found in the nucleus.
(c) Protons and electrons are about the same mass.
(d) The word atom comes from an ancient Greek word that means "indivisible".

Q3. Which of the following was the first person to use the term atom?

- (a) Democritus (b) Ernest Rutherford
(c) Thomson (d) Bohr

Q4. Different models of the atom progressively explained new observations.



Which of the following is the correct order of the development of the atomic model over time, with the earliest first?

(a) b, c, d, e, a

(b) e, d, c, b, a

(c) d, e, c, b, a

(d) b, c, a, d, e

Q5. Which of the following statement is false?

- (a) The electron was discovered in 1897.
(b) The proton was discovered by Ernest Rutherford.
(c) From Rutherford's description the electron cloud is 1000.
(d) The electron number of the atom was proposed by Bohr.

Q6. Which of the following substance particle has a negative charge?

- (a) Proton (b) Neutron (c) Electron

Q7. Which of the following substance particle has a positive charge?

- (a) Electron (b) Neutron (c) Proton

Q8. Which of the following substance particle has the smallest mass?

- (a) Proton (b) Neutron (c) Electron



Which of the following statement is false?

- (a) He represents mass number.
(b) He represents chemical symbol for the element.
(c) He represents atomic number.
(d) The number of neutrons is equal to the total number of He and He.

Q9. For an electrically neutral atom, the number of protons is equal to the number of _____.

- (a) protons (b) electrons (c) ions

Q10. Which is the difference between nucleus of two atoms?



(a) Nucleus A has more protons than nucleus B.

(b) Nucleus A has more neutrons than nucleus B.

(c) Nucleus A has more electrons than nucleus B.

(d) Nucleus A has higher atomic number than nucleus B.

Atomic Theory and Structure of the Atom - Crossword Puzzle

Complete the crossword puzzle below.



Across	Down
1. A scientific description that is comparing two substances or systems.	1. The smallest part of matter that has the chemical and physical properties of the substance.
2. The total number of protons and neutrons in the nucleus of an atom.	2. An electron particle that has a charge.
3. A positively charged particle that is about the size of the atom.	3. The number of protons in the nucleus of an atom.
4. A negatively charged particle that is about the size of the atom.	4. The smallest part of matter that has the chemical and physical properties of the substance.
5. A particle that has a positive charge.	5. A particle that has a negative charge.
6. A particle that has a negative charge.	6. A particle that has a positive charge.
7. A particle that has a positive charge.	7. A particle that has a negative charge.
8. A particle that has a negative charge.	8. A particle that has a positive charge.
9. A particle that has a positive charge.	9. A particle that has a negative charge.
10. A particle that has a negative charge.	10. A particle that has a positive charge.

Atomic Structure Webquest Worksheet Answer Key

NARAYAN CHANGDER



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Atomic Structure Workbook Tina Youngman, 2017

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