

## Regent's Practice Questions

1. Which pair most represents atoms of the same element?
  - A.  $^{12}\text{C}$  and  $^{13}\text{C}$
  - B.  $^{12}\text{C}$  and  $^{12}\text{N}$
  - C.  $^{12}\text{C}$  and  $^{12}\text{O}$
  - D.  $^{12}\text{C}$  and  $^{13}\text{C}$
2. All samples of an element are composed of atoms with the same
  - A. atomic mass
  - B. atomic number
  - C. number of neutrons
  - D. number of protons
3. In a sample of the element phosphorus, each atom has
  - A. 15 protons
  - B. 30 neutrons
  - C. 15 protons and neutrons
  - D. 30 neutrons
4. The atomic number of any atom is equal to the number of
  - A. neutrons in that atom, only
  - B. protons in that atom, only
  - C. neutrons plus protons in that atom
  - D. protons plus electrons in that atom
5. Two atoms will always have the same atomic number if they have the same
  - A. mass number
  - B. number of protons
  - C. number of neutrons
  - D. number of electrons
6. Which is the symbol for an atom containing 20 protons and 23 neutrons?
 

<ol style="list-style-type: none"> <li>A. <math>^{23}\text{Ca}</math></li> <li>B. <math>^{43}\text{Ca}</math></li> </ol>	<ol style="list-style-type: none"> <li>C. <math>^{23}\text{Ca}</math></li> <li>D. <math>^{43}\text{Ca}</math></li> </ol>
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# Atomic Structure Practice Regents Answers

**Effiong Eyo**



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