

## Chapter 22 Assignment Answers – Organic Chemistry Unit

Chapter 22: 37 – 41, 50, 53, 55, 56, 59, 66, 73

37. pentane:  $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$

hexane:  $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$

38. (a) propane                      (b) octane                      (c) pentane

39. Remember... “alkyl” groups are side groups that branch off a longer-chain hydrocarbon.

**methyl** –  $\text{CH}_3$       **ethyl** –  $\text{CH}_2\text{CH}_3$       **propyl** –  $\text{CH}_2\text{CH}_2\text{CH}_3$

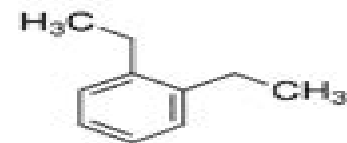
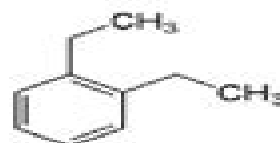
40. (a) 2-methylbutane

(b) 2,3-dimethylbutane

(c) 3-methylhexane

41. Alkane molecules are nonpolar because all of the bonds within the molecule are nonpolar bonds. You *can* make an alkane molecule polar by adding a functional group to it, however.

50. Both structures are the same, because the alternating double bonds don't really exist! Remember that the carbon-to-carbon bonds in benzene are all identical.



# Assessment Chemistry Answers Pg 232

**D Kirk**



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