## Naming Amines

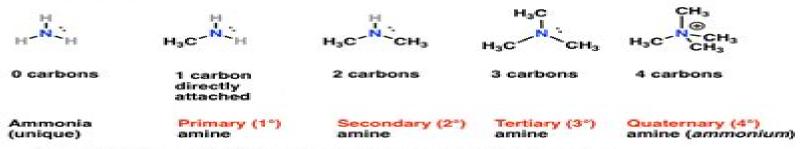
#### Amines

- Amines are organic compounds containing an -NH<sub>2</sub> group.
- They derive from ammonia where an alkyl or aryl group substitutes for one or more hydrogen atoms of NH<sub>3</sub>.

# Functional Group Priorities (Highest → Lowest)

- Carboxylie Acids
- Carboxylic Acid Derivatives
- Nitriles
- Aldehydes
- Ketones
- Alcohols
- Amines
- Alkenes
- Halogenoalkanes

#### Derivatives of Ammonia



Note: A C=N bond will effectively count as 2 carbons bonded to the nitrogen. Note: Quaternary ammonium salts are derived from ammonia but not classed as amines.

#### Naming Rules

- If the amines is the highest priority functional group, use the suffix —amine, e.g. propan-1-amine.
- If there is a higher priority functional group, use the prefix amino-, e.g. 2-aminopropanoic acid.

#### Secondary Amines:

- The longest carbon chain attached to the nitrogen is named as the parent.
- Shorter chains are named as N-substitutions, e.g. N-methylethanamine.

#### Tertiary Amines:

- The longest carbon chain attached to the nitrogen is named as the parent.
- Shorter chains are named as N-substitutions.
- 2 N-prefixes are used to show both N-substitutions.
- The prefixes are named alphabetically, e.g. N-ethyl-N-methylpropan-1-amine.
- N,N-di- is used for identical R-groups, e.g. N,N-diethylphenylamine.

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