

Definitions

Atom =

Particles that make up all substances.

Element =

A substance made up of only one kind of atom.

Compound =

A substance made of different types of atoms joined together.

Mixture =

A substance existing of atoms that can be easily separated as they are not joined together.



Periodic Table of the Elements

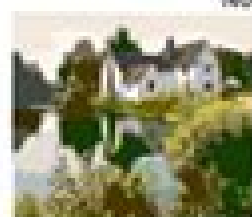
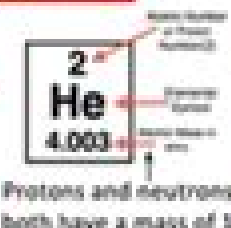
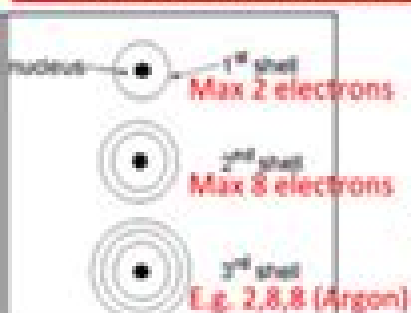
1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
H	He	Li	Be	B	C	N	O	F	Ne	Na	Mg	Al	Si	P	S	Cl	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	

No charge

Group 0 or 8 are **UNREACTIVE**
They have a full outer shell of electrons

Periodic table arranges elements by **ATOMIC NUMBER** (proton number/the small one)

The number of outer shell electrons match the group the element is found in.
E.g. Lithium 2,1 is a group 1 element.



Used for building

LIMESTONE

Calcium Carbonate CaCO_3

HEAT
(thermal decomposition)

Calcium Oxide CaO

Limewater

The test for CO_2

Ca(OH)_2
Or Calcium Hydroxide comes from adding water to CaO

Carbonates Decompose when heated
= Oxide + CO_2
React with acid
= Salt + water + CO_2

Balanced Equations



C=1
H=4
O=4

=

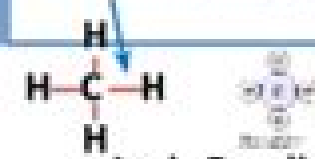
C=1
H=4
O=4

A balanced equation has the same number of atoms of each element on both sides

Using Molecules: In this equation, the large number in front of the chemical symbol tells us the molecules needed.
e.g. 1 CH_4 molecule reacts with 2 O_2 molecules

Covalent Bonding

Non-metals sharing electrons



Ionic Bonding



Good: More jobs
Bad: Destroy habitats and Landscape
Would you want to live next to one?

QUARRYING

Limestone is heated in a Rotary Lime Kiln
You then have Cement or Mortar by adding water and sand.
If you add crushed rocks you get Concrete

Metals - Very useful
e.g. Copper wires conduct well
Most need to be ALLOYED to make them harder

TRANSITION METALS



Metals which are un-reactive are found in their **NATIVE STATE** e.g. GOLD
More reactive metals are found as METALS ORES and need to be EXTRACTED



C1 CHEMISTRY

Aqa Chemistry Specimen Paper Unit 1 Gcse

Joacim Rocklöv



Aqa Chemistry Specimen Paper Unit 1 Gcse:

Reviewing **Aqa Chemistry Specimen Paper Unit 1 Gcse**: Unlocking the Spellbinding Force of Linguistics

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