3.6.2 Acids and alkala



dwiffy the best indicator to distinguish between solidions of different pill, using data provided.

The data and disservations to determine the \$4.0°.

Explain You neutralization way flore you used in a

Describe a method by how to make a recent

Specification

where they are and and alled

a solution and explain what this shows.

range of situations,

Device an employ to compare how well indigention remediat work.

No.

The priof's solution depends on the strength of the acid strong acids have lower phi values than week acids.

Mixing an acid and situal produces a chemical reaction, reutralisation, forming a chemical called a self and sader

Factor

Acquitane a pribeton 7, neutral sciutore have a pH of 7, skale have a pH alone 7.

Adds and alliate can be complete or lettert and require softe handing.

Hydrochure, suiture and other acid are enough acids.

Assets and other and we was acids.

Reserve St.

pM. Scale of acidly and shalledy from \$16.54.

Indicators. Substances used to identify whether unknown solutions are scool or silvatine.

See: A substance that neutralism at acid those that doodne in vater are called shalls.

Concertration A resource of the number of particles in a given solution.

Line

Given the names of an acid and an alsoi, work out the name of the soft produced when they react.

Debute the hazarts of different places and acids using this about their concertration and pin

Extrada the pit of an acid based on internation from nuclions

Lesson 1: Chemical and Physical Reactions

Lesson 2: Acids and Alkalis

Lesson 3: Acids and Metals

Lesson 4: Acids and Carbonates

Lesson 5: Red Cabbage Indicator RP

Lesson 6: Universal Indicator

Lesson 7: Neutralisation

Lesson Sequence

Pupils should have been taught to: va-

- compare and group materials together, according to whether they are solids, liquids or gases
- observe that some materials change state when they are heated or cooled

YES

- compare and group together everyday materials on the basis of their properties.
- explain that some changes result in the formation of new materials, and that this kind of change is not usually reversible, including the action of acid on bicarbonate of soda

KS2 Links

From AQA GCSE Combined Science Chemistry:

5.4.2.1 Reactions of acids with metals

5.4.2.2 Neutralisation of acids and salt production

5.4.2.3 Soluble salts

5.4.2.4 The pH scale and neutralisation

5.4.2.5 Strong and weak acids (HT only)

From AQA GCSE Chemistry:

4.4.2.1 Reactions of acids with metals

4.4.2.2 Neutralisation of acids and salt production

4.4.2.3 Soluble salts.

4.4.2.4 The pH scale and neutralisation

4.4.2.5 Titrations (chemistry only)

4.4.2.6 Strong and weak acids (HT enly)

KS4 Links

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Lawrie Ryan

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