



14 Organic analysis

- 1 (a) (i) O—H (alcohols) [1]
(ii) C=O [1]
(b) $\text{CH}_3\text{CH}_2\text{OH} + [\text{O}] \rightarrow \text{CH}_3\text{CHO} + \text{H}_2\text{O}$ [1]
(c) There is no O—H bond in ethanal [1]
(d) Ethanal is being oxidised to form ethanoic acid [1]; there is an O—H group in ethanoic acid [1]
- 2 (a) There is no O—H (alcohols) stretch between 3230 and 3550 cm^{-1} [1]
(b) Cyclohexene [1] C_6H_{10} , because there is a C=C absorption between 1620 and 1680 cm^{-1} [1]
(c) A precise value of m/z for the molecular ion is formed that can identify the compound [1]
- 3 Alcohol OH absorption in different place ($3230\text{--}3550\text{ cm}^{-1}$) from acid OH absorption ($2500\text{--}3000\text{ cm}^{-1}$) [1]
The C=O in acids has an absorption at $1680\text{--}1750\text{ cm}^{-1}$ [1]

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Oxford International AQA Examinations: International A Level Chemistry Ted Lister, Janet Renshaw, 2016-10-13

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