

ACID & BASE

- Provide optimal cell function
 - Cells secrete **acid** and **bases** to maintain proper pH for enzymes to work.
- They both provide homeostatic regulation of pH of the body's ECF (extracellular fluid)
 - The Carbonic Acid-Bicarb Buffer System = pH homeostasis of blood

ACID

CO₂ is regulated by the lungs + contributes to acid

- found in the stomach + urine + kidneys *
 - ex: acid reflux, BPH, kidney failure

BASE

Bicarb HCO₃ = form of carbon dioxide in our blood

- found in intestines *
 - Regulated, excreted + reabsorbed in kidneys *
 - Ex C-diff → bacteria in colon

2 FUNCTIONS

Metabolic + Respiratory

METABOLIC

primary change is the **concentration of bicarbonate.**

METABOLIC **ACIDOSIS**

bicarb ↓ r/t the amount of acid in the body

- Increased Hydrogen ions
 - CAUSES: kidney failure, DKA, lactic acidosis (shock, sepsis), severe diarrhea
 - put pt. on bicarb drip to increase levels
- Respiratory compensation
 - Increased RR to rid of acid

METABOLIC **ALKALOSIS**

excess bicarb r/t the amount of acid in the body

- Decreased Hydrogen ions
 - CAUSES: loss of gastric fluids, antacid overdose
 - ex: too much NGT suctioning, hypokalemia, vomiting

RESPIRATORY

primary change is the **concentration of carbonic acid**

RESPIRATORY **ACIDOSIS**

CO₂ is retained + inadequate O₂ exchange → increases amount of carbonic acid in the body

- CAUSES: Hypoventilation
- ex: Asthma, COPD, Pneumonia
- Decreased RR
 - Sleep apnea, drug overdose, head injuries

Compensation?

- Increasing RR

RESPIRATORY **ALKALOSIS**

Too much CO₂ is lost + carbonic acid levels fall

- Pt has increased RR
 - ex: hyperventilation due to anxiety attack

Compensation?

- slow + shallow breaths

Acid Base Study Guide Answer Key

Susan McMurry, John McMurry



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Table of Contents Acid Base Study Guide Answer Key

1. Understanding the eBook Acid Base Study Guide Answer Key
 - The Rise of Digital Reading Acid Base Study Guide Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Acid Base Study Guide Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Acid Base Study Guide Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Acid Base Study Guide Answer Key
 - Personalized Recommendations
 - Acid Base Study Guide Answer Key User Reviews and Ratings
 - Acid Base Study Guide Answer Key and Bestseller Lists
5. Accessing Acid Base Study Guide Answer Key Free and Paid eBooks
 - Acid Base Study Guide Answer Key Public Domain eBooks
 - Acid Base Study Guide Answer Key eBook Subscription Services
 - Acid Base Study Guide Answer Key Budget-Friendly Options
6. Navigating Acid Base Study Guide Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Acid Base Study Guide Answer Key Compatibility with Devices
 - Acid Base Study Guide Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Acid Base Study Guide Answer Key
 - Highlighting and Note-Taking Acid Base Study Guide Answer Key
 - Interactive Elements Acid Base Study Guide Answer Key
8. Staying Engaged with Acid Base Study Guide Answer Key

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Acid Base Study Guide Answer Key
9. Balancing eBooks and Physical Books Acid Base Study Guide Answer Key
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Acid Base Study Guide Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Acid Base Study Guide Answer Key
- Setting Reading Goals Acid Base Study Guide Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Acid Base Study Guide Answer Key
- Fact-Checking eBook Content of Acid Base Study Guide Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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