Student Name:				Student ID:		
CHE	M 2218	Inorganic	Chemistry I	(Final Exa	ım sample paper)	
Part !	I: Multiple	e Choices (2)	points each, 50	points in total)		
1.	For a cube, the number of C_6 , C_4 , and C_3 rotational axes is (a) 1, 3, and 4, respectively (b) 0, 2, and 4, respectively (c) 0, 3, and 4, respective					
2.	The CO diatomic molecule contains infinite number of (a) σ_v (b) i (c) σ_b (d) C_2					
3. I	dentify the w groups.		the following Doon, D _{2v} , C _i , I,		presentations of point	
4.	The Fe ₂ (CO (a) i) ₉ molecule sho (b) σ	wn does not po (c) S ₃	sses what elem (d) C ₂	ent of symmetry?	
5.	A molecule having a point group of D _{4h} contains all the symmetry elements of a molecule having a point group of (a) D _{4d} (b) D _{2d} (c) D _{3h} (d) none of the above					
6.	How many (a) 6	mirror planes of (b) 9	symmetry are (c) 12	there in an icos (d) 15	sahedron? (e) 18	
7.		nmetry label for ture (you are giv			ted with a trigonal	
8.	Identify the H ₂ Se vs. H ₂ :	stronger acid? S [Al(E	I ₂ O) ₆] ³⁺ vs. [Fe	$(H_2O)_6]^{3+}$	00	
9.		n the following) ₆] ³⁺ (b) [/			(d) Mo(CO) ₆	
10.	Predict which way the following reactions will go (left or right) (a) $\{AlF_6\}^{3*} + Ga(CN)_6\}^{3*} \rightarrow \{GaF_6\}^{3*} + Al(CN)_6\}^{3*}$ (a) $W(CO)_5(PMe_3) + NMe_3 \rightarrow W(CO)_5(NMe_3) + PMe_3$					
11.		rongest base tov (b) NHMe ₂		(d) PMe ₃		
12.	(a) H + OH (b) PF ₃ + F ₂			the acid-base	reactions?	

(d) Mo + 6CO → Mo(CO)6-

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