

19

ACIDS, BASES, AND SALTS

SECTION 19.1 ACID-BASE THEORIES (pages 587–593)

This section compares and contrasts acids and bases as defined by the theories of Arrhenius, Brønsted-Lowry, and Lewis. It also identifies conjugate acid-base pairs in acid-base reactions.

► Properties of Acids and Bases (pages 587–588)

1. Circle the letters of all the terms that complete the sentence correctly.
The properties of acids include _____.
 - a. reacting with metals to produce oxygen
 - b. giving foods a sour taste
 - c. forming solutions that conduct electricity
 - d. causing indicators to change color
2. Bases are compounds that react with acids to form _____ and a(n) _____.
3. Circle the letters of all the terms that complete the sentence correctly.
The properties of bases include _____.
 - a. tasting bitter
 - b. feeling slippery
 - c. changing the color of an indicator
 - d. always acting as a strong electrolyte

► Arrhenius Acids and Bases (pages 588–590)

4. Match the number of ionizable hydrogens with the type of acid.

| | |
|-------------|---------------|
| _____ one | a. diprotic |
| _____ two | b. triprotic |
| _____ three | c. monoprotic |
5. Is the following sentence true or false? Only the hydrogens in weak polar bonds are ionizable. _____
6. Hydrogen is joined to a very _____ element in a very polar bond.
7. Alkali metals react with water to produce _____ solutions.

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R.P. Rana



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