

Name: _____

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Nuclear Fission and Fusion

Learning Objectives

- Compare and contrast nuclear fission and nuclear fusion.
- Understand the conservation laws that apply to nuclear reactions.

Vocabulary

- Fission, fusion, transmutation and nuclide.

Why?

Fission and fusion are two processes that alter the nucleus of an atom. Nuclear fission provides the energy in nuclear power plants and fusion is the source of the sun's energy. The use of fission in power plants can help conserve fossil fuels. Without the energy produced by the fusion of the hydrogen in the sun, the Earth would quickly change into a cold planet that could not support life as we know it.

Information

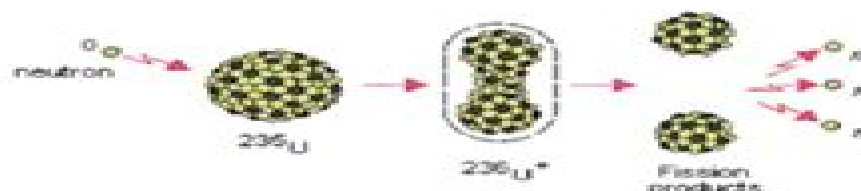
Transmutation is the transformation of the nucleus of an atom so that the atom is changed from one element into a different element. This can be accomplished through many types of reactions, including fission and fusion.

Nuclear transformations always obey two fundamental conservation laws: (1) mass number is conserved and (2) electrical charge is conserved. Energy and mass are not served, but can be inter-converted according to Einstein's equation, $E=mc^2$. The amount of binding energy released from the nucleus of an atom during fission and fusion is so tremendous that a small fraction of mass is converted to energy; this is referred to as the mass defect.

Model

1. Fission

The process of fission occurs when a nucleus splits into smaller pieces. Fission can be induced by a nucleus capturing slow moving neutrons, which results in the nucleus becoming very unstable.



(<http://chemed.chem.purdue.edu/grochessa/topiccon/inner/typ/ch23/Fission.html>)

The following equations represent fission reactions, where n = neutron.



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Ralph Thomas Myers



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1. Understanding the eBook 53 Fission And Fusion Of Atomic Nuclei Answer Key
 - The Rise of Digital Reading 53 Fission And Fusion Of Atomic Nuclei Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying 53 Fission And Fusion Of Atomic Nuclei Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 53 Fission And Fusion Of Atomic Nuclei Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from 53 Fission And Fusion Of Atomic Nuclei Answer Key
 - Personalized Recommendations
 - 53 Fission And Fusion Of Atomic Nuclei Answer Key User Reviews and Ratings
 - 53 Fission And Fusion Of Atomic Nuclei Answer Key and Bestseller Lists
5. Accessing 53 Fission And Fusion Of Atomic Nuclei Answer Key Free and Paid eBooks
 - 53 Fission And Fusion Of Atomic Nuclei Answer Key Public Domain eBooks
 - 53 Fission And Fusion Of Atomic Nuclei Answer Key eBook Subscription Services
 - 53 Fission And Fusion Of Atomic Nuclei Answer Key Budget-Friendly Options
6. Navigating 53 Fission And Fusion Of Atomic Nuclei Answer Key eBook Formats

- ePub, PDF, MOBI, and More
 - 53 Fission And Fusion Of Atomic Nuclei Answer Key Compatibility with Devices
 - 53 Fission And Fusion Of Atomic Nuclei Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 53 Fission And Fusion Of Atomic Nuclei Answer Key
 - Highlighting and Note-Taking 53 Fission And Fusion Of Atomic Nuclei Answer Key
 - Interactive Elements 53 Fission And Fusion Of Atomic Nuclei Answer Key
 8. Staying Engaged with 53 Fission And Fusion Of Atomic Nuclei Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 53 Fission And Fusion Of Atomic Nuclei Answer Key
 9. Balancing eBooks and Physical Books 53 Fission And Fusion Of Atomic Nuclei Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 53 Fission And Fusion Of Atomic Nuclei Answer Key
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine 53 Fission And Fusion Of Atomic Nuclei Answer Key
 - Setting Reading Goals 53 Fission And Fusion Of Atomic Nuclei Answer Key
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of 53 Fission And Fusion Of Atomic Nuclei Answer Key
 - Fact-Checking eBook Content of 53 Fission And Fusion Of Atomic Nuclei Answer Key
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
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