

Naming and Writing Formulas for Covalent (Molecular) Compounds

Name: _____

Part A: Name the covalent compounds.

1. CO carbon monoxide
2. CO₂ carbon dioxide
3. N₂O₃ dinitrogen trioxide
4. NP nitrogen monophosphide
5. SCl₂ sulfur dichloride
6. P₂O₅ diphosphorus pentoxide
7. NBr₃ nitrogen tribromide
8. Cl₄ chlorine tetrachloride
9. CCl₄ carbon tetrachloride
10. PF₅ phosphorus pentafluoride
11. SeF₂ selenium difluoride
12. TeBr₂ tellurium dibromide
13. P₂S₅ diphosphorus pentasulfide
14. C₃N₄ tricarbon tetranitrogen
15. CH₄ carbon tetrahydride
16. PH₃ phosphorus trihydride

Part B: Write the Chemical Formula

1. carbon tetrafluoride CF₄
2. silicon dioxide SiO₂
3. dinitrogen trisulfide N₂S₃
4. phosphorus mononitride PN
5. carbon disulfide CS₂
6. dinitrogen trichloride N₂Cl₃
7. silicon tetrabromide SiBr₄
8. carbon dioxide CO₂
9. nitrogen trifluoride NF₃
10. boron trisulfide BS₃
11. disulfur trioxide S₂O₃
12. selenium tetrafluoride SeF₄
13. diphosphorus pentasulfide P₂S₅
14. sulfur dibromide SBr₂
15. carbon tetrachloride CCl₄
16. dinitrogen tetroxide N₂O₄

Building And Naming Molecules Lab Notes Answer Key

Lingjun Ying



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In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Building And Naming Molecules Lab Notes Answer Key**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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