

## BrainPOP: Atomic Model

Using \_\_\_\_\_ means, scientists have been able to construct different models of the \_\_\_\_\_. The atomic model has taken years and years to build and people still tweak it now and again, hopefully making it more accurate.

**1802**- At the beginning of the 19<sup>th</sup> century, Englishman \_\_\_\_\_ built an atomic model called the "\_\_\_\_\_. According to his theory, each type of \_\_\_\_\_ is made of only one type of \_\_\_\_\_. Also each type of atom could be put together into a group called an \_\_\_\_\_ - and every atom of a given element was \_\_\_\_\_. For example, gold atoms make \_\_\_\_\_ and iron atoms make \_\_\_\_\_, and so on. Dalton also believed that atoms could not be \_\_\_\_\_, divided into smaller parts, or \_\_\_\_\_.

**1897**- About 100 years after Dalton, British scientists \_\_\_\_\_ discovered the \_\_\_\_\_. Thompson stated that negatively charged electrons must be stuck all throughout a positively charged area. Think of it like the atom is cookie dough and the \_\_\_\_\_ are chocolate chips. The discovery of electrons was a big deal. But it left some big questions- what were the positive particles holding those electrons in place?

**1909**- In 1909, a New Zealand born British scientist named \_\_\_\_\_ and his team set out to answer some of these questions. After performing in depth experiments using gold foil, Rutherford determined that a gold atom must have some very small \_\_\_\_\_ charged mass. He named this mass the \_\_\_\_\_. He called the positively charged nucleus particles \_\_\_\_\_, and said that the \_\_\_\_\_ were scattered in \_\_\_\_\_ around the nucleus.

**James Chadwick**, a student of Rutherford's, found new particles in the nucleus that weren't affected by an electric field at all. Chadwick called these uncharged particles \_\_\_\_\_.

Also in the early 20<sup>th</sup> century, scientists discovered that electrons are arranged in energy levels. The lower energy levels are closer to the \_\_\_\_\_ and can hold just a few \_\_\_\_\_. The higher energy levels are farther from the nucleus and can hold more \_\_\_\_\_. Electrons are so small and \_\_\_\_\_ that their energy levels are not neat planet-like orbits around the nucleus. Instead, they exist in a region called the electron cloud. The atomic model is constantly evolving as scientists discover more and more about what makes the universe tick.

# Brainpop Chemical Bonds Answer Key

**Sebastian Brünink**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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## **Table of Contents Brainpop Chemical Bonds Answer Key**

1. Understanding the eBook Brainpop Chemical Bonds Answer Key
  - The Rise of Digital Reading Brainpop Chemical Bonds Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Brainpop Chemical Bonds Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Brainpop Chemical Bonds Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Brainpop Chemical Bonds Answer Key

- Personalized Recommendations
- Brainpop Chemical Bonds Answer Key User Reviews and Ratings
- Brainpop Chemical Bonds Answer Key and Bestseller Lists
- 5. Accessing Brainpop Chemical Bonds Answer Key Free and Paid eBooks
  - Brainpop Chemical Bonds Answer Key Public Domain eBooks
  - Brainpop Chemical Bonds Answer Key eBook Subscription Services
  - Brainpop Chemical Bonds Answer Key Budget-Friendly Options
- 6. Navigating Brainpop Chemical Bonds Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Brainpop Chemical Bonds Answer Key Compatibility with Devices
  - Brainpop Chemical Bonds Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Brainpop Chemical Bonds Answer Key
  - Highlighting and Note-Taking Brainpop Chemical Bonds Answer Key
  - Interactive Elements Brainpop Chemical Bonds Answer Key
- 8. Staying Engaged with Brainpop Chemical Bonds Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Brainpop Chemical Bonds Answer Key
- 9. Balancing eBooks and Physical Books Brainpop Chemical Bonds Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Brainpop Chemical Bonds Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Brainpop Chemical Bonds Answer Key
  - Setting Reading Goals Brainpop Chemical Bonds Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Brainpop Chemical Bonds Answer Key

- Fact-Checking eBook Content of Brainpop Chemical Bonds Answer Key
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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