#### UNIVERSITY OF BOTSWANA

## 2021 / 2022 (GENERAL CHEMISTRY II)

## SEMESTER II - FINAL EXAMINATION

#### FRONT PAGE

COURSE NO: CHE 102

DURATION: 2 HRS

DATE: 17th May 2022

TITLE PAPER: GENERAL CHEMISTRY

SUBJECT: CHEMISTRY

TITLE OF EXAMINATION: BSc Level 100

LAST NAME:	INITIALS:	INITIALS:		
ID#	LAB	LAB		
GROUP (A, B) If not taking lab, write N/A	DAY:	TIME:		

#### WRITE ALL ANSWERS ON THIS QUESTION PAPER

Specific heat capacity of H<sub>2</sub>O (I) = 4.184 J/(°C.g) 1 atm = 760 Torr Avogadro constant = 6.022 x 10<sup>23</sup> mol<sup>-1</sup> R = 8.314 J. K<sup>-1</sup>, mol<sup>-1</sup> = 0.08206 L, atm. K<sup>-1</sup>, mol<sup>-1</sup> 1L<sup>-</sup> atm = 101.33 J

The paper has two parts. For Section A (multiple-choice questions), circle the letter for the correct answer for each question.

For questions in Section B, use the space provided to write your answers.

IMPORTANT: It is the student's responsibility to report if any page is missing in this paper. The paper has 20 multiple choice questions in Section A and 4 questions in Section B.

## NOTE: Questions are printed on BOTH sides of the paper.

## Do not write in this table.

MC	1	2	3	4	Total	%
60	10	08	10	12	100	

Number of pages: 12 including the periodic table

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