

Molecular Structure

Instructions: Fill in each section of this worksheet and submit it to your instructor for grading.

Review Questions

1. Write a statement explaining which characteristics of an atom determine the VSEPR shape of a molecule (hint: look at the VSEPR geometry chart).
VSEPR shows the shape of a molecule and the arrangement of its electron pairs and the atom, it predicts the lone pairs and it predicts shape.
2. How does the force between the atoms impact the shape of a molecule?
The force between the atom determines the shape.

Activity: First row is done for you as a model.

Compound	Draw Lewis Structure	Use VSEPR Chart to determine:	Determine the structures of the molecule	Draw the VSEPR model
H ₂ O		Total Domains: 4 Bonded Atoms: 2 Lone Pairs: 2	Molecular Shape: Bent Electron Geometry: Tetrahedral	
SiH ₄		Total Domains: 4 Bonded Atoms: 4 Lone Pairs: 0	Molecular Shape: tetrahedral Electron Geometry: tetrahedral	
Br ₂		Total Domains: 1 Bonded Atoms: 1 Lone Pairs: 0	Molecular Shape: linear Electron Geometry: linear	

Bonding And Molecular Structure Worksheet Answers

Year 9

M Woodhall



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