

# Biology SOL Review Packet

Name: \_\_\_\_\_

## I. Scientific Investigation:

### A. *Steps to the Scientific Method*

1. **Make observations/Do research:** to determine what problem you want to address
2. **Develop a Research Question:** what specifically do you want to determine?
3. **Develop a hypothesis:** based on research from a variety of sources
  - a. **scientific journals:** are the best place to locate current findings on the newest technologies
  - b. **encyclopedias:** are a good place to find information on extinct species or historical theories
  - c. **state/local agencies:** can help with local policies or local research (example: can help research the effects of pesticides on the squirrel population)
4. **Conduct a Controlled Experiment-** testing the hypothesis and collecting data and observations
5. **Interpret and Analyze Data:** using tables and graphs
6. **Draw Conclusions-** including improvements for future experiments
7. **Communicate Results:** so others can build on the experiment

**B. *Terms: Word Bank:*** *experiment, variables, hypothesis, constants, control, independent variable, dependent variable*

1. \_\_\_\_\_ - is an educated guess/prediction; usually in "IF...THEN" form.
2. \_\_\_\_\_ - are the factors that are measured in an experiment.
3. \_\_\_\_\_ - is the variable that you purposely change...variable "I" change.
4. \_\_\_\_\_ - is the variable that changes as a result of changing the IV.
5. \_\_\_\_\_ - is the baseline measurement that you compare your data to.
6. \_\_\_\_\_ - the thing(s) that are purposely kept the same in the experiment.
7. \_\_\_\_\_ - is a structured way to test a hypothesis.

# Biology Sol Review Packet Goochland Answer Key

**Ensheng Dong**



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web october 3 2017 abstract the techniques used in this lab are quantitative transfer and vacuum filtration with the reaction of 8 001 grams of copper ii sulfate  $\text{CuSO}_4$  and 2 0153 grams of iron powder  $\text{Fe}$  the goal of this experiment was to determine the product of copper ii sulfate with iron

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

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web through this experiment we sought to use stoichiometric principles which derive from stoichiometry the study of substance quantities in ratio or proportion in a chemical reaction in this lab experiment the principles help distinguish the equation between iron powder and aqueous copper ii sulfate in the iron copper ii sulfate redox reaction

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web this experiment is focused on a reaction between iron metal and a solution of copper ii sulfate we are able to figure out the chemical equation between the two by using stoichiometric principles the iron displaces the copper ii sulfate ions the

copper ii is then reduced and gains electrons

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web stoichiometry can be defined as the relationship between the relative quantities of substances taking part in a reaction or forming a compound stoichiometric principles can be used to deduce the appropriate chemical equation for a reaction in our experiment we used solid iron in the form of powder was added to a solution of copper ii sulfate

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web under the finalization of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away copper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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