



1. Carbon and chlorine atoms **join** to form salt molecules.
 - A. join and chlorine atoms
 - B. ice melts to form water
 - C. Carbon dioxide freezes to form dry ice
 - D. Salt and water mix to form salt water
2. In the equation $2H_2 + O_2 \rightarrow 2H_2O$, what are the reactants?
 - A. Hydrogen atoms
 - B. Hydrogen and oxygen molecules
 - C. Water molecules
 - D. Ice crystals
3. Sulfuric acid is made of two hydrogen atoms (H), one sulfur atom (S), and four oxygen atoms (O). What is its molecular formula?
 - A. $2H_4SO_2$
 - B. H_4SO_2
 - C. H_2SO_4
 - D. H_2SO_4
4. What is the best synonym for the word "stoichiometry"?
 - A. Counting atoms
 - B. Balancing chemical equations
 - C. Discovering new chemical formulas
 - D. Breaking the chemical bonds that hold molecules together
5. The chemical symbol "W" most likely represents:
 - A. Carbon

- reactants?
 - A. The products usually weigh more than the reactants
 - B. The products often have completely different properties than the reactants
 - C. The products usually have more atoms than the reactants
 - D. The products are usually more toxic than the reactants
7. What's the problem with this chemical equation? $H_2 + O_2 \rightarrow H_2O$
 - A. There's a missing hydrogen in the reactants
 - B. There's a missing oxygen in the product
 - C. There's an additional oxygen in the product
 - D. There's an additional hydrogen in the reactants
8. What's the easiest way to balance a chemical equation?
 - A. Trial and error
 - B. Using the periodic table
 - C. Complex algebra
 - D. Calculus
9. Which reactant is missing from the following equation? $P + PCl_5 \rightarrow PCl_3 + PCl_5$
 - A. One hydrogen atom
 - B. One phosphorus atom
 - C. Three hydrogen atoms
 - D. One oxygen atom
10. What is the product of the following equation? $2H_2 + O_2 \rightarrow$



Handwritten blue notes: "2" and "O" with a "2" next to it.

Brainpop Chemical Equations Quiz Answers

Heather Hattori, Richard H. Langley

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the authors' names.

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