

## BrainPOP: Atomic Model

Using \_\_\_\_\_ means, scientists have been able to construct different models of the \_\_\_\_\_. The atomic model has taken years and years to build and people still tweak it now and again, hopefully making it more accurate.

**1802**- At the beginning of the 19<sup>th</sup> century, Englishman \_\_\_\_\_ built an atomic model called the "\_\_\_\_\_. According to his theory, each type of \_\_\_\_\_ is made of only one type of \_\_\_\_\_. Also each type of atom could be put together into a group called an \_\_\_\_\_ - and every atom of a given element was \_\_\_\_\_. For example, gold atoms make \_\_\_\_\_ and iron atoms make \_\_\_\_\_, and so on. Dalton also believed that atoms could not be \_\_\_\_\_, divided into smaller parts, or \_\_\_\_\_.

**1897**- About 100 years after Dalton, British scientists \_\_\_\_\_ discovered the \_\_\_\_\_. Thompson stated that negatively charged electrons must be stuck all throughout a positively charged area. Think of it like the atom is cookie dough and the \_\_\_\_\_ are chocolate chips. The discovery of electrons was a big deal. But it left some big questions- what were the positive particles holding those electrons in place?

**1909**- In 1909, a New Zealand born British scientist named \_\_\_\_\_ and his team set out to answer some of these questions. After performing in depth experiments using gold foil, Rutherford determined that a gold atom must have some very small \_\_\_\_\_ charged mass. He named this mass the \_\_\_\_\_. He called the positively charged nucleus particles \_\_\_\_\_, and said that the \_\_\_\_\_ were scattered in \_\_\_\_\_ around the nucleus.

**James Chadwick**, a student of Rutherford's, found new particles in the nucleus that weren't affected by an electric field at all. Chadwick called these uncharged particles \_\_\_\_\_.

Also in the early 20<sup>th</sup> century, scientists discovered that electrons are arranged in energy levels. The lower energy levels are closer to the \_\_\_\_\_ and can hold just a few \_\_\_\_\_. The higher energy levels are farther from the nucleus and can hold more \_\_\_\_\_. Electrons are so small and \_\_\_\_\_ that their energy levels are not neat planet-like orbits around the nucleus. Instead, they exist in a region called the electron cloud. The atomic model is constantly evolving as scientists discover more and more about what makes the universe tick.

# Brainpop Activity Science And Chemistry Bonds Answer Key

**Victor M. Corman**



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