Ans: a) Gold sol

7. What is the unit of rate constant for zero order reaction?

a) Mol L⁻¹s⁻¹

b) Mol*L s*

c) s. 1

d) Mol⁻¹ L⁻¹s⁻¹

(1)

Ans: Mol L's'

 "The limiting molar conductivity of an electrolyte is obtained by adding the limiting molar conductivities of cations and anions of the electrolyte". Name the law.

Ans: a) Kohlrausch's law

Calculate the no. of tetrahedral voids, if Mg crystal contains 'n' number of atoms. (1)

Ans. No. of tetrahedral voids = 2n

Part I SECTION B

(Answer any Four questions from 10 to 13, each question carries 1 scores - non focus area)

10. Which of these is temperature dependent?

a) Molarity b) Molality c) Mole fraction

d) ppm

(II)

Ans: a) Molarity

11. What is the basicity of H₃PO₃?

a) I

b) 2

c) 3

d) 4

(1)

Ans: b) 2

12. CH₃-CH₂-CH₂-Br ^{alc. KOH} → ?

(II)

Ans: $CH_3 - CH = CH_2 \{Propene\}$

13. $----+ CH_3MgBr \longrightarrow CH_3-CH_2-OMgBr \xrightarrow{H2O} CH_3-CH_2-OH + MgBr(OH)$ (1)

Ans: HCHO

Part II SECTION A

(Answer any Two questions from 14 to 17, each question carries 2 scores - focus area)

Bonar Law Chapter 11 Chemistry Answerkey

J Elliott

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