Name : _______ Date: ______

Classifying Chemical Reactions

- A. Classify the following reaction as synthesis, decomposition, single-replacement, double-replacement, and combustion:
- 1) Al₂(SO₄)₃ + 3 BaCl₂ --- 3 BaSO₄ + 2 AlCl₃ _____
- 2) Al₂S₃ -- 2 Al + 2 S
- 3) 2 NaOH + CuSO₄ --- Na₂SO₄ + Cu(OH)₂ _____
- 4) Fe + H₂SO₄ → FeSO₄ + H₂ _____
- 5) C₄H₁₂ + 7O₂ 4CO₂ + 6H₂O _____
- B. Balance the following equations and indicate the type of reaction:
- 1) H_2S + O_2 \longrightarrow SO_2 + H_2O _____
- 2) C₅H₉O + O₂ → CO₂ + H₂O _____
- 3) Al + NiBr₂ → AlBr₃ + Ni _____
- 4) AI + O₂ → AI₂O₃ _____
- 5) H₂O₂ + O₂ ______

Blank Classifying Chemical Reactions

Guillaume Favre

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