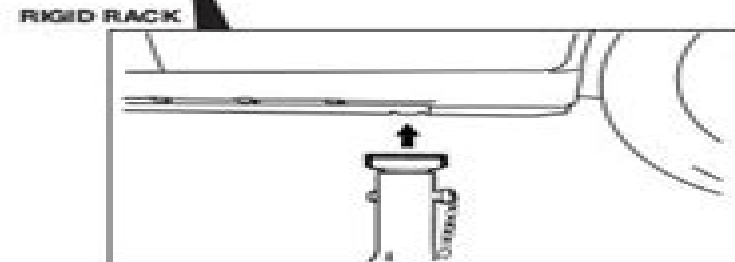
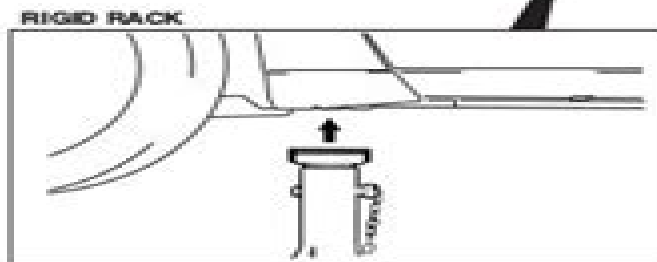
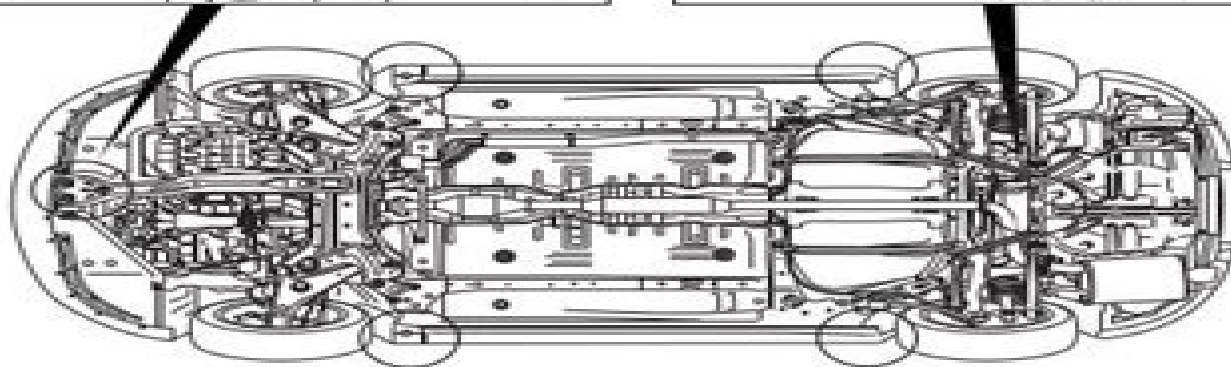
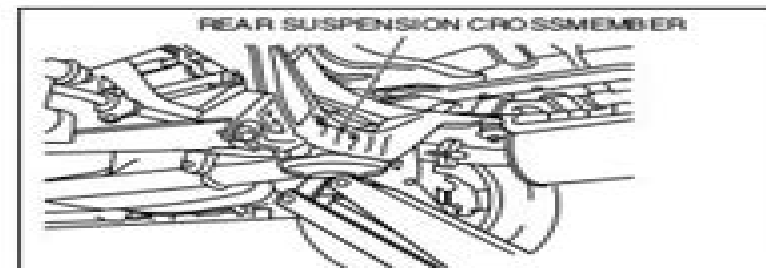
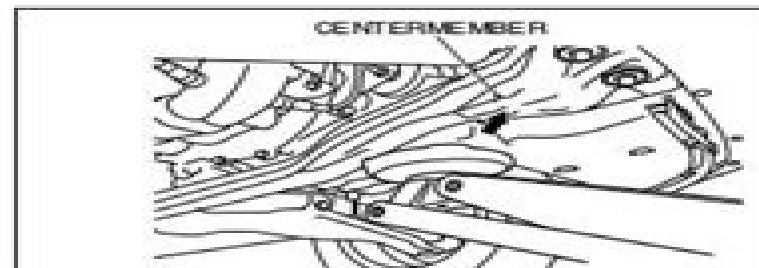


**LIFTING, JACKING SUPPORT LOCATION
FLOOR JACK****⚠ CAUTION**

- Never place a support at any point other than the specified one, or that point will be deformed.
- For lifting, put rubber or similar material between the side sill and rigid rack, otherwise the side sill area will be damaged.



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 0.2074 g of calcium hydroxide Ca(OH)_2 in 40.00 ml of solution

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 b 4.25 g of NH_3 in 0.500 l of solution the concentration of NH_3 in household ammonia

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 molality = $\frac{\text{moles solute}}{\text{kilograms solvent}}$
 mathematical manipulation of molality is the

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web 1 molarity mv mass molar mass $\times 0.4500 \text{ l} = 0.825 \text{ g}$
 $\frac{0.825 \text{ g}}{141.9579 \text{ g/mol}} \times 0.0129 \text{ m} = 0.000581158 \text{ mol}$
 $\frac{0.000581158 \text{ mol}}{0.4500 \text{ kg}} = 0.00129 \text{ m}$
 3 mole fraction Na_2HPO_4
 $\frac{0.825 \text{ g}}{141.9579 \text{ g/mol}} = 0.00581158 \text{ mol}$
 $\frac{0.00581158 \text{ mol}}{0.4500 \text{ g}} = 0.0129 \text{ mol/kg}$
 $\frac{0.0129 \text{ mol}}{0.4500 \text{ kg}} = 0.0287 \text{ mol/kg}$

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$$\frac{\text{mass of a component}}{\text{mass of component in solution total mass of solution}} \times 100$$
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