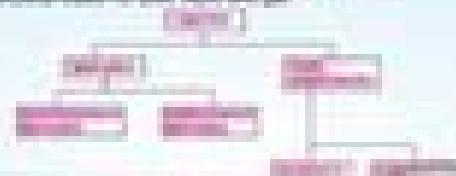


1. CLASSIFICATION OF MATTER

- Matter can exist in three physical states: solid, liquid and gas.



- Various properties of matter are used to differentiate between the two.
- Matter is classified into pure and impure.
- Homogeneous mixture: composition is uniform throughout.
- Heterogeneous mixture: composition is not uniform.
- Solution is a homogeneous mixture.
- Suspension is a heterogeneous mixture.
- Colloid is a heterogeneous mixture.

2. PROPERTIES OF MATTER AND THEIR MEASUREMENT

- Properties of matter are physical and chemical properties.
- Physical properties are those which can be measured without changing the identity of the substance.
- Chemical properties are those which can be measured by changing the identity of the substance.
- Colour, odour, melting point, density, etc. are physical properties.
- Combustion, corrosion, etc. are chemical properties.
- Quantitative measurement of physical properties is called measurement.
- Measurement is done by using standard units.
- Measurement is done by using standard units.
- Measurement is done by using standard units.

UNIT OF MEASUREMENT OF PHYSICAL PROPERTIES

Length (m)	Mass (kg)	Time (s)	Temperature (K)	Amount of substance (mol)
1000 m = 1 km	1000 g = 1 kg	1000 s = 1 min	1000 K = 1 °C	1000 mol = 1 kmol

UNIT OF MEASUREMENT OF PHYSICAL PROPERTIES

- 1. 1000 m = 1 km, 1000 g = 1 kg, 1000 s = 1 min, 1000 K = 1 °C, 1000 mol = 1 kmol

3. SIGNIFICANT FIGURES

Significant figures are those digits which are known with certainty, plus one additional uncertain digit.

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