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1. Re-read your notes and use your Cornell-questions to quiz yourself.
2. Re-read pages 40 – 73 of your textbook.
3. Be able to define the key vocabulary from the chapter.
4. Review your Reading Checks and Work Book, making sure all is complete.
5. Be able to answer the questions below. They are a guide for your studying.
6. Complete the online quizzes at <http://www.bcs-science.com/bc9/>

• elements	• atomic number	• period	• electron shells
• compounds	• atomic mass	• chemical family	• valence electrons
• chemical symbol	• atomic mass unit	• alkali metals	• valence shell
• chemical properties	• ion charge	• alkaline earth metals	• noble gas stability
• metals	• ion	• halogens	
• non-metals	• multiple ion charge	• noble gases	
• periodic table	• metalloids	• Bohr model of the atom	

1. How many elements are there?
2. Give 2 examples of elements at each of the following states:
 - a). solid
 - b). liquid
 - c). gas
3. Most of the elements names are derived from what languages?
4. Be familiar with the physical and chemical properties, some uses, and interesting information of the following elements:
 - a). Hydrogen
 - b). Iron
 - c). Oxygen
 - d). Sodium
 - e). Chlorine
 - f). Mercury
 - g). Silver
 - h). Silicon
5. Complete the Section 2.1 Quiz online at <http://www.b-c-science.com/bc9/>

- Who is Dmitri Mendeleev?
 - Describe his early work on organizing the elements.
 - Describe Mendeleev's special insight in the organization of elements.
- What does the Periodic Table show for each element?
- Differentiate atomic number from atomic mass and choose a specific element for example.
 - Describe the pattern of each on the Periodic Table.
- Differentiate atoms from ions.
- What are the 2 types of ions?

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Clemens Wendtner



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