

ATOMIC STRUCTURE QUIZ



Multiple Choice: Read the question carefully and select the choice that best answers the question. 2 pt each

- 1.) The charge of a proton is....
 - a.) Negative
 - b.) Positive
 - c.) Neutral
 - d.) Unknown
- 2.) What does the atomic number tell you?
 - a.) Number of neutrons
 - b.) The size of the atom
 - c.) The mass of the atom
 - d.) The number of protons
- 3.) How can you find the mass of an isotope?
 - a.) Adding protons and neutrons
 - b.) Adding protons and electrons
 - c.) Subtracting protons and neutrons
 - d.) Subtracting electrons and neutrons
- 4.) What is the Atomic Mass of Lithium?
 - a.) 6.94
 - b.) 3
 - c.) 4
 - d.) 6.023
- 5.) The charge of an electron is....
 - a.) Negative
 - b.) Positive
 - c.) Neutral
 - d.) Unknown
- 6.) In a Bohr model, how many electrons can go in the first orbital?
 - a.) 8
 - b.) 16
 - c.) 2
 - d.) 3
- 7.) What is the element symbol for Radon?
 - a.) Rb
 - b.) Ra
 - c.) Rn
 - d.) Ru
- 8.) Which subatomic particle does not have a mass?
 - a.) Neutr
 - b.) Proton
 - c.) Neutron
 - d.) Electron
- 9.) In a neutral atom, the number electrons equals the number of....
 - a.) Neutrons
 - b.) Protons
 - c.) Mass
 - d.) Cannot Determine
- 10.) What is the name of the element with the symbol, Hg
 - a.) Mercury
 - b.) Bromine
 - c.) Iodine
 - d.) Helium

Atomic Structure Test Questions

Yan Bai



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