

FELONY - SEXUAL ABUSE (1901-1910)



1. Re-read your notes and use your Cornell-questions to quiz yourself.
2. Re-read pages 40 – 73 of your textbook.
3. Be able to define the key vocabulary from the chapter.
4. Review your Reading Checks and Work Book, making sure all is complete.
5. Be able to answer the questions below. They are a guide for your studying.
6. Complete the online quizzes at <http://www.bcs-science.com/bc9/>

• elements	• atomic number	• period	• electron shells
• compounds	• atomic mass	• chemical family	• valence electrons
• chemical symbol	• atomic mass unit	• alkali metals	• valence shell
• chemical properties	• ion charge	• alkaline earth metals	• noble gas stability
• metals	• ion	• halogens	
• non-metals	• multiple ion charge	• noble gases	
• periodic table	• metalloids	• Bohr model of the atom	

1. How many elements are there?
2. Give 2 examples of elements at each of the following states:
 - a). solid
 - b). liquid
 - c). gas
3. Most of the elements names are derived from what languages?
4. Be familiar with the physical and chemical properties, some uses, and interesting information of the following elements:
 - a). Hydrogen
 - b). Iron
 - c). Oxygen
 - d). Sodium
 - e). Chlorine
 - f). Mercury
 - g). Silver
 - h). Silicon
5. Complete the Section 2.1 Quiz online at <http://www.b-c-science.com/b-c9/>

- Who is Dmitri Mendeleev?
 - Describe his early work on organizing the elements.
 - Describe Mendeleev's special insight in the organization of elements.
- What does the Periodic Table show for each element?
- Differentiate atomic number from atomic mass and choose a specific element for example.
 - Describe the pattern of each on the Periodic Table.
- Differentiate atoms from ions.
- What are the 2 types of ions?

Bc Science 9 Chemistry Study Guide

William Allen Miller



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serpentine is a large mineral group with hydrous magnesium iron silicate that forms from the serpentinization of rocks like peridotite and olivine as a silicate crystal it forms many varieties of green brown blue red black white pink orange and yellow colors with a greasy silky soapy or flaky texture

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serpentinization is a hydration and metamorphic transformation of ferromagnesian minerals such as olivine and pyroxene in mafic and ultramafic rock to produce serpentinite 1 minerals formed by serpentinization include the serpentine group minerals antigorite lizardite chrysotile brucite talc ni fe alloys and magnetite

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of characteristic of or resembling a serpent as in form or movement having a winding course as a road sinuous shrewd wily or cunning noun a device on a harquebus lock for holding the

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serpentinite is a rock composed predominantly of one or more serpentine group minerals the name originating from the similarity of the texture of the rock to that of the skin of a snake serpentinite has been called serpentine or serpentine rock particularly in older geological texts and in wider cultural settings

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serpentine any of a group of hydrous magnesium rich silicate minerals the composition of these common rock forming minerals approximates $\text{mg}_3\text{si}_2\text{o}_5\text{oh}_4$ serpentinite generally occurs in three polymorphs chrysotile a fibrous variety used as asbestos antigorite a variety occurring in either

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