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Research Article

Crop Residue Ashes as Adsorbents for Basic Dye (Methylene Blue) Removal: Adsorption Kinetics and Dynamics

In the quest for identifying low-cost, locally available and effective adsorbents, pigeon pea (Cejanus cejan L.) residue ash (PRA) and sunflower (Helianthus annus L.) residue ash (SRA) were evaluated as adsorbents for sequestering a basic dive, methylene blue (MB). from agueous solution. Effects of contact time, initial solution pH and adsorbe at dose on MB adsorption were investigated. MB adsorption onto PRA and SRA was dependent on initial solution pH, with adsorption being greater at higher pH. Percent MB removal increased with increase in adsorbent's dose. Adsorption followed pseudo-second order kinetics. Intraparticle diffusion was not the sole rate-controlling step. Both film diffusion and intraparticle diffusion controlled the adsorption process. The equilibrium data fitted well to Langmuir isotherm as compared to Freundlich isotherm, on the basis of coefficient of determination and error analysis. The maximum MB adsorption capacities of PRA and SRA were found as 58.8 and 62.5 mgg 1, respectively. The free emergy change and dimensionless equilibrium parameter revealed spontaneous and favorable nature of MB adsorption onto crop residue ashes. Results demonstrate that the PRA and SRA have high adsorptive potential as compared to many previously reported ash based adsorbents, and hence can be used for treating water contaminated with cationic dyes.

Keywords: Adsorption kinetics; Grop residue ash; Dye removal; Isotherm models

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1 Introduction

Industries such as teorific, rubber, paper, leather, plastics, cosmetic, printing, etc., use a wide variety of frynthetic dyes to produce a myriad of products to satisfy dynamic and ever evolving consumer's demand. About 5–20% of dyes used in these industries are, however, lost in industrial wastewater [1, 2]. Through the disposal of untreated dyscontaining industrial wastewater to streams, dyes enter aquatic entirements and pose a threat to the aquatic occupied not only by altering the physicochemical properties or biological activities of water [3], but also from toxicological stand point [4]. Through water bodies, dyes may gain entry to food chains and pose serious threat to human health [5, 6]. Proper treatment of dyscontaminated water is, therefore, an important issue for environment friendly disposal of stained wasterwater.

As a result of extensive research work, a number of dye removal techniques such as advorption, biosception, biodegratation, electrodhemical degradation, photo-dhemical degradation, oragulation, floorulation, socioated degradation, membrane fibration, etc., are

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Abbreviations: PTIE. Fourier wandoms IE; MB. Methylene blue: OC. organic carbon: plltque phi of acto point charge:: PEA, pigeon pea residue shi: SEA, surficewer residue shi: TOC, total OC.

available for treating stained wastewater [7]. Of all these techniques, dipe removal through adsorption is most effective and practically less bliefs, 9]. Activated carbon is an unadoushtedly effective adsorbers, but its high cost limits its large-scale usage, particularly in developing countries. In order to circumvent these problems, there is a growing interest in the use of low-cost and easily available adsorbers, including various natural materials, wastes/hyproducts from agriculture or industry, for treatment of dye-contaminated wastewater [1, 10, 11]. However, not much research attention is paid to evaluate the potential of cop residue ash, the secondary by-product, as adsorbents to remove due.

Coop residues are generated in huge quantities as a primary byproduct of intensive agricultural production system. For example, in India about 523Mt per annum crop residue is generated [12]. Cropresidues of low lignocellulosic composition are used as a resource either in manure/compost preparation or as cartle feed. But, substantial quantities of high lignocellulosic residues of crops such as pigeon pea (9Mt per annum in India) and sunflower (1Mt per annum in India) are neither used for feeding cartle nor preparing manure B.2. Keeping these large volumes of high lignocellulosic crop residues in field often cause obstruction for agreemachinery/ implements and also harbors pests. Burning of these crop residues, either in situ as farmer's common practice or off the field as fuels for thermal energy purposes generates a large quantity of cropresidue ash, the secondary by-product. For which there is no alternative use.

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- 1. Understanding the eBook Ashes No More Releasing The Residue And Embracing The Strength
 - The Rise of Digital Reading Ashes No More Releasing The Residue And Embracing The Strength
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Ashes No More Releasing The Residue And Embracing The Strength
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ashes No More Releasing The Residue And Embracing The Strength
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Ashes No More Releasing The Residue And Embracing The Strength
 - Personalized Recommendations
 - Ashes No More Releasing The Residue And Embracing The Strength User Reviews and Ratings
 - Ashes No More Releasing The Residue And Embracing The Strength and Bestseller Lists
- 5. Accessing Ashes No More Releasing The Residue And Embracing The Strength Free and Paid eBooks

- Ashes No More Releasing The Residue And Embracing The Strength Public Domain eBooks
- Ashes No More Releasing The Residue And Embracing The Strength eBook Subscription Services
- Ashes No More Releasing The Residue And Embracing The Strength Budget-Friendly Options
- 6. Navigating Ashes No More Releasing The Residue And Embracing The Strength eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Ashes No More Releasing The Residue And Embracing The Strength Compatibility with Devices
 - Ashes No More Releasing The Residue And Embracing The Strength Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ashes No More Releasing The Residue And Embracing The Strength
 - Highlighting and Note-Taking Ashes No More Releasing The Residue And Embracing The Strength
 - Interactive Elements Ashes No More Releasing The Residue And Embracing The Strength
- 8. Staying Engaged with Ashes No More Releasing The Residue And Embracing The Strength
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ashes No More Releasing The Residue And Embracing The Strength
- 9. Balancing eBooks and Physical Books Ashes No More Releasing The Residue And Embracing The Strength
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ashes No More Releasing The Residue And Embracing The Strength
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ashes No More Releasing The Residue And Embracing The Strength
 - Setting Reading Goals Ashes No More Releasing The Residue And Embracing The Strength
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ashes No More Releasing The Residue And Embracing The Strength
 - Fact-Checking eBook Content of Ashes No More Releasing The Residue And Embracing The Strength
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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