Study Buddy: Atomic Structure and Nuclear Chemistry

Scientist	Experiment	Findings	Limitations	Model
Democritus	No experiment	Atoms are the smallest form of matter and cannot be created, destroyed or divided	Did not explain chemical behavior	
Dalton	Use experiments of previous scientists	Postulates: Atoms of elements are identical; atoms can't be created or destroyed; atoms combine in whole number ratios in a chemical reaction	Did not include sub- atomic particles	
Thomson	Cathode ray	Discovered the electron "Plum- pudding" Model	Did not explain deflection pattern of alpha particles as seen by Rutherford	
Rutherford	Gold foil	Discovered the proton and nucleus. Atoms contain mostly empty space.	Could not explain why electrons do not crash into the nucleus	0
Chadwick	Alpha particles	Discovered the neutron	No modification of atom model	No new model/neutron later add to nucleus
Bohr	Uses findings from previous scientists and Quantum theory	Electrons are in orbits of differing energy levels.	Does not explain the behavior of other elements just hydrogen	(3)
Modern Atom	Multiple experiments	Electron cloud and dual nature of electrons	Still doesn't explain the exceptions	· ===

Parts of an Atom (8th Grade Review)

Sub-atomic Particle	Location	Charge	Mass
Proton (p*)	Nucleus	1+	1
Neutron (n*)	Nucleus	0	1
Electron (er)	outside the nucleus	1-	0 or 1/1840

Nuclear Symbol Carbon-14

Protons:	- 6	
Neutrons:	-8	
Atomic Number:	6	
Mass Number:	14	

Identify the proton, electron, neutron in the model below. electron neutron proton What is the atomic number of the above models? 2 What is the atomic mass? 4.00 amu What element is this? Helium

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JE Gale

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Study Material of Electrical Power Systems for GATE & ESE (Theory & Questions) Volume 1 Umesh Dhande, 2024-11-07 This comprehensive guide is designed to cater to the growing demand for accurate and concise concepts and formulas for electrical engineering power systems Vol 1 The book s key features include 1 Step by Step Solutions Detailed easy to follow solutions to all questions 2 Chapter Wise and Year Wise Analysis In depth analysis of questions organized by chapter and year 3 Detailed Explanations Clear explanations of each question ensuring a thorough understanding of the concepts 4 Simple and Easy to Understand Language Solutions are presented in a straightforward and *General Chemistry Workbook* Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection accessible manner of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final **Nuclear Science Abstracts** ,1975 Research in Education ,1973 exam covering the entire material Basic Concepts of Chemistry, Study Guide Leo J. Malone, 1989-02-21 This Third Edition revised to Education .1973-05 provide smoother transitions between topics employs a concise yet informal approach to basic chemistry organized to help students employ basic math skills and problem solving strategies Writing style is straightforward and presentation incorporates many concrete analogies to clarify new concepts Includes many illustrative worked examples Atoms for Peace Manual United States. Congress. Senate. Committee on Foreign Relations, United States, 1955 **Study Guide and Student Solutions Manual** Douglas Brandt, 2000 Physics for Scientists and Engineers combines outstanding pedagogy with a clear and direct narrative and applications that draw the reader into the physics. The new edition features an unrivaled suite of media and on line resources that enhance the understanding of physics Many new topics have been incorporated such as the Otto cycle lens combinations three phase alternating current and many more New developments and discoveries in physics have been added including the Hubble space telescope age and inflation of the universe and distant planets Modern physics topics are often discussed within the framework of classical physics where appropriate For scientists and engineers who are interested in learning physics **Energy and Water Development Appropriations for 2000: Secretary of Energy, Department of Administration** United States. Congress. House. Committee on Appropriations. Subcommittee on Energy and Water Development, 1999 **Energy and Water Development Appropriations for 2000** United States. Congress. House. Committee on Appropriations. Subcommittee on Energy and Water Development, 1999 Participation Act for the International Atomic Energy Agency United States. Congress. Joint Committee on Atomic Energy. Special Subcommittee on Outer Space Propulsion, United States. Congress. Joint Committee on Atomic Energy.

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