

EASEL ACID AND BASE QUIZ

Multiple Choice: Select the most correct answer. (15 K)

1. Which of the following indicates an acidic solution?

- A) pH = 7.5 B) pH = 7 C) pH = 6.5
D) more than one correct response E) no correct response

2. Which of the following statements about weak acids is correct?

- A) Weak acids always contain C atoms
B) A percentage dissociation of 50% is a weak acid
C) Weak acid molecules readily lose their acidic hydrogen atoms
D) more than one correct response
E) none of the above

3. Which of the following equations represents an acid-base neutralization reaction?

- A) $\text{H}_2\text{SO}_4 + \text{Zn} \rightarrow \text{ZnSO}_4 + \text{H}_2$ B) $\text{HNO}_3 + \text{KOH} \rightarrow \text{KNO}_3 + \text{H}_2\text{O}$
C) $\text{Ba}(\text{OH})_2 + \text{Na}_2\text{SO}_4 \rightarrow \text{BaSO}_4 + 2\text{NaOH}$ D) more than one correct response
E) none of the above

4. If the pH of a solution increases from 4.0 to 6.0, the hydronium ion concentration

- A) increases by a factor of 1.5. B) increases by a factor of 100.
C) decreases by a factor of 1.5. D) decreases by a factor of 100.

5. The most basic oxides are formed from elements found in the _____ region of the periodic table.

- A) right B) left C) center D) top E) none of the above

6. When $\text{CaO}(\text{s})$ reacts with water which of the following is true?

- a) The solution contains $\text{OH}^-_{(\text{aq})}$ and $\text{Ca}^{2+}_{(\text{aq})}$ b) The solution contains $\text{O}^{2-}_{(\text{aq})}$ and $\text{Ca}^{2+}_{(\text{aq})}$
c) The solution contains $\text{O}^{2-}_{(\text{aq})}$, $\text{OH}^-_{(\text{aq})}$, and $\text{Ca}^{2+}_{(\text{aq})}$ d) The solution contains $\text{CaO}_{(\text{aq})}$

7. A weak base:

- A) partially dissociates in water to give a little $\text{OH}^-_{(\text{aq})}$
B) fully reacts with water to form $\text{OH}^-_{(\text{aq})}$
C) completely dissociates to form $\text{OH}^-_{(\text{aq})}$
D) has a very high pH
E) contains a group 1 metal and a hydroxide

8. Which of the following is NOT a property of bases?

- A) They react with active metals. B) They taste bitter and feel slippery
C) They have a pH greater than 7. D) They turn red litmus blue E) They neutralize acids.

Acids And Bases Test A Answers

Sandra F. Smith, Smith



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Joseph A. Mascetta, Mark Kernion, 2020-12-01 Note College Board has discontinued the SAT Subject Tests in the US. The tests will be available outside the US in June 2021 and then be discontinued. Barron's SAT Subject Test Chemistry with 7 Practice Tests features in-depth review of all topics on the exam and full-length practice tests in the book and online. This edition includes one full-length diagnostic test to help you assess your strengths and weaknesses. Comprehensive review of all topics on the exam including introductory chemistry, atomic structure and the periodic table, bonding, chemical formulas, gases and laws, stoichiometry, liquids, solids and phase changes, chemical reactions and thermochemistry, chemical reactions, chemical equilibrium, acids, bases and salts, oxidation-reduction, carbon and organic chemistry and the laboratory. Four full-length practice tests that reflect the actual SAT Subject Test Chemistry exam in length, question types and degree of difficulty. Two full-length online practice tests with answer explanations and automated scoring. Appendices which include the periodic table, important equation constants and data tables and a glossary of chemistry terms. **Saunders Comprehensive Review for the NCLEX-RN® Examination, Fourth South Asia Edition - E-Book** Annu Kaushik, 2022-11-15

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Senyuk, Darrell D. Ebbing, 1984 **MSCEIS 2019** Lala Septem Riza, Eka Cahya Prima, Toni Hadibarata, Peter John Aubusson, 2020-07-30 The 7th Mathematics Science and Computer Science Education International Seminar MSCEIS was held by the Faculty of Mathematics and Natural Science Education Universitas Pendidikan Indonesia UPI and the collaboration with 12 University associated in Asosiasi MIPA LPTK Indonesia AMLI consisting of Universitas Negeri Semarang UNNES Universitas Pendidikan Indonesia UPI Universitas Negeri Yogyakarta UNY Universitas Negeri Malang UM Universitas Negeri Jakarta UNJ Universitas Negeri Medan UNIMED Universitas Negeri Padang UNP Universitas Negeri Manado UNIMA Universitas Negeri Makassar UNM Universitas Pendidikan Ganesha UNDHIKSA Universitas Negeri Gorontalo UNG and Universitas Negeri Surabaya UNESA. In this year MSCEIS 2019 takes the following theme: Mathematics

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