Chemistry Final Exam Question and Answer

which type of reaction does this diagram represent?

Answer: nuclear fusion because atomic nuclei combine to form heavier nucleus.

4/2 He + 4/2 He > ?
if this equation was completed, which statement would it best support?
Answer: nuclear fusion produces elements that are heavier than helium.

a fusion reaction takes place between carbon and another element, neutrons are released, and a different element is formed, the different element is? Answer: heavier than helium.

which best describes a chain reaction associated with a nuclear reaction?

Answer: neutrons released during a fission reaction causes other nuclei to split.

which radioisotope is used as a tracer in nuclear scanning? Answer: technetium-99m

uranium-232 has a half life of 68.8 years, after 344.0 years how much uranium will remain from a 100.0g sample? Answer: 3.13 g

a sample contains 16.75 g of radioisotopes U-236 and 50.25 g of its daughter isotope, Th-233, how long did it take for decay to take place if one half life of U-236 is 23 million years?

Answer: 46 million years

an article about half lives describes a daughter isotope, what is a daughter isotope? Answer: the isotope that forms from the radioactive decay of a less stable isotope

a radioactive nuclide has 53 protons and 78 neutrons, which is the symbol of this nuclide?

Answer: 131/53 I

which represents a balanced nuclear equation? Answer: 24/11 Na > 24/12 Mg + 0/-1 e

consider the nuclear equation below. 239/94 Pu > X + 4/2 He what is X? Answer: 235/92 U

the symbol for a radioactive nuclide is 210/84 Po. which statement is correct? Answer: the number of neutrons present in the nucleus of each atom is 126.

examine the nuclear reaction 1/1 H + 1/0 n > 2/1 H. why is this classified as a nuclear reaction rather than a chemical reaction?

Answer: a change has occurred in a nucleus.

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