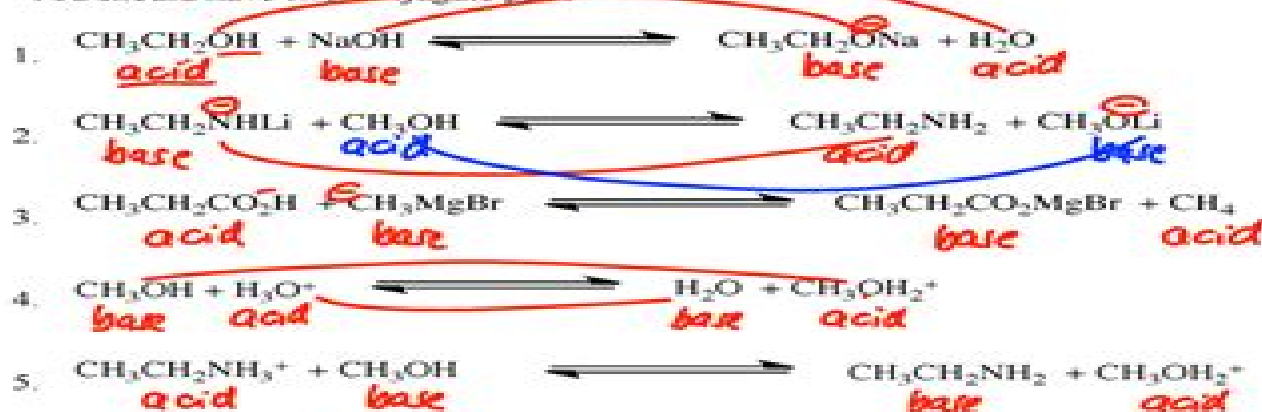


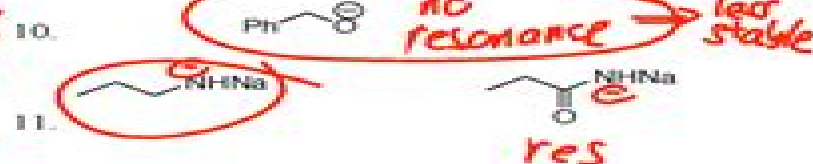
A. Identify each chemical as either an "acid" or a "base" in the following reactions, and identify "conjugate" relationships.

- You should have one acid and one base on each side

- You should have two conjugate pairs



B. Choose the More Basic for Each of the Following Pairs (Single Variable). You can use stability to decide.



Base: less stable $\Rightarrow$ more reactive

Keys:

1. Charge
2. Electronegativity
3. Resonance

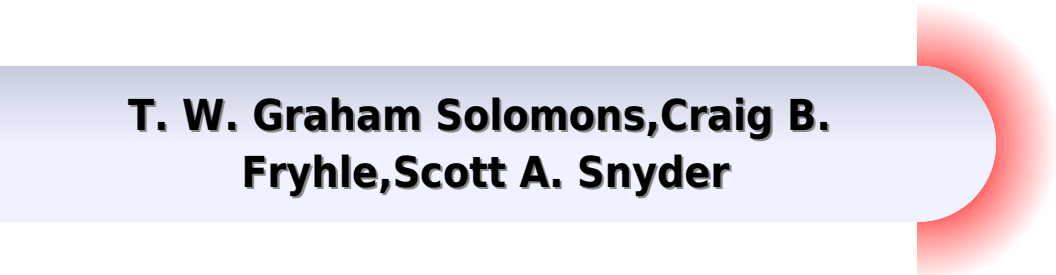
Base
 ① charge: anion more basic than neutral

② eneg: C^- N^- O^- F^-
 Stability $\rightarrow$

③ res anion stabilized $\Rightarrow$ less basic
 res $\Rightarrow$ stabilizes

Acids And Bases Practice Problems Answers

**T. W. Graham Solomons, Craig B.
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Laboratory Manual S. Suzanne Nielsen, 2017-06-07 This third edition laboratory manual was written to accompany Food Analysis Fifth Edition by the same author New to this third edition of the laboratory manual are four introductory chapters that complement both the textbook chapters and the laboratory exercises The 24 laboratory exercises in the manual cover 21 of the 35 chapters in the textbook Many of the laboratory exercises have multiple sections to cover several methods of analysis for a particular food component or characteristic Most of the laboratory exercises include the following background reading assignment objective principle of method chemicals reagents precautions and waste disposal supplies equipment procedure data and calculations questions and references This laboratory manual is ideal for the laboratory portion of undergraduate courses in food analysis *Organic Chemistry* T. W. Graham Solomons, Craig B. Fryhle, Scott A.

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