

ent 3

er Mg (0.10 mol) was mixed with 0.50 mol of
r and added to 100 mL of H_2O at $20^\circ C$ in an insu-
lainer. The mixture was stirred constantly and the
temperature increase that occurred within
a recorded. The procedure was repeated with dif-
ferent amounts of NaCl (see Table 2).

NaCl

Table 2

NaCl added (mol)	Maximum temperature increase ($^\circ C$)
0	0
0.025	19
0.050	

33. Which of the following factors affecting the reaction of Mg and H_2O was studied in Experiment 1, but not in Experiments 2 or 3?

- A. Reaction temperature
- B. Addition of NaCl
- C. Addition of Fe
- D. Surface area of Mg

34. An engineer is designing an FRP to be used with 100 mL of H_2O , that will most rapidly generate the greatest amount of heat with the least amount of materials. Based on the results of Experiments 1, 2, and 3, which of the following specifications should the engineer choose?

- F. 0.50 mol Fe powder, 0.125 mol Mg powder
- G. 0.50 mol Fe powder, 0.125 mol Mg ribbon
- H. 0.70 mol Fe powder, 0.125 mol Mg powder
- J. 0.70 mol Fe powder, 0.125 mol Mg ribbon

35. It has been observed that $Mg(OH)_2$ (magnesium hydroxide) forms an unreactive coating on the surface of Mg. Which of the following modifications to the reaction is most consistent with the results of the experiments?

- A. NaCl absorbs heat from the reaction, so the reaction is less exothermic.
- B. NaCl binds with the $Mg(OH)_2$, so the coating is less effective.
- C. NaCl reacts with H_2O to form a solution that is more acidic than H_2O .

Act 59f Practice Answer Key Science

Michal Rosen-Zvi

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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1. Understanding the eBook Act 59f Practice Answer Key Science
 - The Rise of Digital Reading Act 59f Practice Answer Key Science
 - Advantages of eBooks Over Traditional Books
2. Identifying Act 59f Practice Answer Key Science
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Act 59f Practice Answer Key Science
 - User-Friendly Interface
4. Exploring eBook Recommendations from Act 59f Practice Answer Key Science
 - Personalized Recommendations
 - Act 59f Practice Answer Key Science User Reviews and Ratings
 - Act 59f Practice Answer Key Science and Bestseller Lists
5. Accessing Act 59f Practice Answer Key Science Free and Paid eBooks
 - Act 59f Practice Answer Key Science Public Domain eBooks
 - Act 59f Practice Answer Key Science eBook Subscription Services
 - Act 59f Practice Answer Key Science Budget-Friendly Options
6. Navigating Act 59f Practice Answer Key Science eBook Formats

- ePub, PDF, MOBI, and More
- Act 59f Practice Answer Key Science Compatibility with Devices
- Act 59f Practice Answer Key Science Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Act 59f Practice Answer Key Science
 - Highlighting and Note-Taking Act 59f Practice Answer Key Science
 - Interactive Elements Act 59f Practice Answer Key Science
- 8. Staying Engaged with Act 59f Practice Answer Key Science
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Act 59f Practice Answer Key Science
- 9. Balancing eBooks and Physical Books Act 59f Practice Answer Key Science
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Act 59f Practice Answer Key Science
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Act 59f Practice Answer Key Science
 - Setting Reading Goals Act 59f Practice Answer Key Science
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Act 59f Practice Answer Key Science
 - Fact-Checking eBook Content of Act 59f Practice Answer Key Science
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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