

Covalent Bonding

High School Chemistry Version

x10

COVALENT BONDING

KEY IDEA

COVALENT compounds

COVALENT BOND

is a bond between two atoms

COVALENT COMPOUND

is a substance that contains covalent bonds



Overlapping orbitals where electrons are shared

HELPS TOGETHER BY THE ELECTROSTATIC ATTRACT

How covalent bonding works



Chlorine has 7 valence electrons



Carbon has 4 valence electrons

they combine to form

This forms



Shared one dot

MOLECULES & Lewis structures

MOLECULE

LEWIS STRUCTURE



Polar & Nonpolar Molecules

have partial charges caused by unequal sharing of electrons

do not have partial charges



covalent properties

COVALENT PROPERTIES ARE MUCH MORE VARIOUS THAN IONIC



between molecules so they are easier to separate

which is also polar, dissolves in water, does not

Visual Guided Notes

Covalent Bonding Guided

Disha Experts

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the 'Disha Experts' text.

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