

ChemThink: Covalent Bonding

Covalent bonding forms when atoms are sharing electrons.

When two atoms get close enough, the nucleus attracts the other atom's
 1. Electrons 2. Neutrons 3. Protons

Before bonding, the atom's electrons spend most of their time
orbiting the nuclei of each atom. Once bonded, the
 electrons spend most of their time between the two nuclei.

Atoms must be able to hold onto their own electrons while
pulling attraction other atom's electrons.

Covalent bonds form between two nonmetals.

When atoms move closer, the potential energy increases decreases.
 At a certain point the potential energy increases. If you try
 and force the atoms closer against the protons in each
 nucleus are repelling each other.

The ideal distance between the atoms is known as the bond length.

Lower in energy = more stable.

Bond Type	Draw an Example	# of paired e ⁻	Total # of e ⁻ shared	Strongest Weakest
Single	$H-H$	1	2	<u>weakest</u>
Double	$O=O$	2	4	—
Triple	$N \equiv N$	3	6	<u>strongest</u>

Covalent Bonding Answer Key For Chemthink

Primrose Kitten

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

Covalent Bonding Answer Key For Chemthink:

Atoms, Electrons, Structure and Bonding Primrose Kitten, 2017-06-19 I'm constantly telling you the best way to learn is by practicing questions so I've made you a book full of practice questions Multiple choice questions to reflect the style of exam questions activities to complete equations for you to balance compounds for you to work out the formula for lots of things that you need to recall and practice long answer exam style questions This book is not designed as a text book or revision guide but as a workbook There are lots of good and bad expensive and free revision guides out there on my YouTube channel and other great websites So there is no point in me adding to the masses All the teaching all the new content is available for free on my YouTube channel this book is for you to practice and learn The best way to approach this is to watch the teaching video and make notes or after class try a section and check the answers Any corrections that are needed after the book is published will be listed on my website www.primrosekitten.com these will be corrected in the next version of the book Answers are provided for the sections where you need to work out the answers for yourself not the sections where you are just filling things in from a video or website

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Some of this content has also been published in the Summer Start to A Level Chemistry and a practice exam paper

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